

Sarcosine, N-(4-chlorobenzoyl)-, isobutyl ester

Inchi:	InChI=1S/C14H18ClNO3/c1-10(2)9-19-13(17)8-16(3)14(18)11-4-6-12(15)7-5-11/h4-7,10
InchiKey:	BIRRTJZKUPXLRD-UHFFFAOYSA-N
Formula:	C14H18ClNO3
SMILES:	CC(C)COC(=O)CN(C)C(=O)c1ccc(Cl)cc1
Mol. weight [g/mol]:	283.75

Physical Properties

Property code	Value	Unit	Source
gf	-96.65	kJ/mol	Joback Method
hf	-418.10	kJ/mol	Joback Method
hfus	33.75	kJ/mol	Joback Method
hvap	71.64	kJ/mol	Joback Method
log10ws	-3.01		Crippen Method
logp	2.611		Crippen Method
mcvol	215.590	ml/mol	McGowan Method
pc	2127.56	kPa	Joback Method
rinpol	2117.00		NIST Webbook
rinpol	2117.00		NIST Webbook
tb	730.97	K	Joback Method
tc	943.15	K	Joback Method
tf	455.96	K	Joback Method
vc	0.802	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	579.70	J/mol×K	730.97	Joback Method
cpg	593.74	J/mol×K	766.33	Joback Method
cpg	606.80	J/mol×K	801.70	Joback Method
cpg	618.91	J/mol×K	837.06	Joback Method
cpg	630.10	J/mol×K	872.43	Joback Method
cpg	640.41	J/mol×K	907.79	Joback Method
cpg	649.88	J/mol×K	943.15	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U321348&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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