

# Benzenamine, N-ethyl-N-nitroso-

<b>Other names:</b>	Aniline, N-ethyl-N-nitroso-EthylNitrosoaniline N-Ethyl-N-nitrosoaniline N-Nitroso-N-ethylaniline Nitrosoethylaniline Nitrosoethylphenylamine NEA N-Ethyl-N-phenylnitrosamine N-Nitrosoethylphenylamine NSC 405070
<b>Inchi:</b>	InChI=1S/C8H10N2O/c1-2-10(9-11)8-6-4-3-5-7-8/h3-7H,2H2,1H3
<b>InchiKey:</b>	WXR XVZXYLBWKR G-UHFFFAOYSA-N
<b>Formula:</b>	C8H10N2O
<b>SMILES:</b>	CCN(N=O)c1ccccc1
<b>Mol. weight [g/mol]:</b>	150.18
<b>CAS:</b>	612-64-6

## Physical Properties

Property code	Value	Unit	Source
chs	-4679.80	kJ/mol	NIST Webbook
hf	-72.58	kJ/mol	Joback Method
h vap	46.82	kJ/mol	Joback Method
log10ws	-2.56		Crippen Method
logp	2.194		Crippen Method
m cvol	121.350	ml/mol	McGowan Method
pc	3564.27	kPa	Joback Method
tb	484.96	K	Joback Method
tc	689.94	K	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C612646&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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