

# Sebacic acid, di(2-isopropylphenyl) ester

**Inchi:** InChI=1S/C28H38O4/c1-21(2)23-15-11-13-17-25(23)31-27(29)19-9-7-5-6-8-10-20-28(30)  
**InchiKey:** WENIXBHBMWHHJJ-UHFFFAOYSA-N  
**Formula:** C28H38O4  
**SMILES:** CC(C)c1cccc1OC(=O)CCCCCCCC(=O)Oc1cccc1C(C)C  
**Mol. weight [g/mol]:** 438.60

## Physical Properties

Property code	Value	Unit	Source
gf	-82.28	kJ/mol	Joback Method
hf	-671.29	kJ/mol	Joback Method
hfus	54.11	kJ/mol	Joback Method
hvap	101.33	kJ/mol	Joback Method
log10ws	-8.63		Crippen Method
logp	7.565		Crippen Method
mcvol	372.740	ml/mol	McGowan Method
pc	997.65	kPa	Joback Method
rinqol	3274.00		NIST Webbook
tb	1055.06	K	Joback Method
tc	1291.84	K	Joback Method
tf	597.52	K	Joback Method
vc	1.423	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1259.91	J/molxK	1055.06	Joback Method
cpg	1316.05	J/molxK	1252.38	Joback Method
cpg	1307.88	J/molxK	1212.91	Joback Method
cpg	1298.24	J/molxK	1173.45	Joback Method
cpg	1287.09	J/molxK	1133.99	Joback Method
cpg	1274.33	J/molxK	1094.52	Joback Method
cpg	1322.85	J/molxK	1291.84	Joback Method
dvisc	0.0000148	Paxs	1055.06	Joback Method
dvisc	0.0000194	Paxs	978.80	Joback Method

dvisc	0.0000268	Paxs	902.55	Joback Method
dvisc	0.0000392	Paxs	826.29	Joback Method
dvisc	0.0000620	Paxs	750.03	Joback Method
dvisc	0.0001088	Paxs	673.78	Joback Method
dvisc	0.0002202	Paxs	597.52	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U354833&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U354833&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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