

# 2-Bromobenzyl alcohol, n-pentyl ether

<b>Inchi:</b>	InChI=1S/C12H17BrO/c1-2-3-6-9-14-10-11-7-4-5-8-12(11)13/h4-5,7-8H,2-3,6,9-10H2,1H
<b>InchiKey:</b>	WJWANQRTZJQLOE-UHFFFAOYSA-N
<b>Formula:</b>	C12H17BrO
<b>SMILES:</b>	CCCCCOCc1ccccc1Br
<b>Mol. weight [g/mol]:</b>	257.17

## Physical Properties

Property code	Value	Unit	Source
gf	62.26	kJ/mol	Joback Method
hf	-171.84	kJ/mol	Joback Method
hfus	26.96	kJ/mol	Joback Method
hvap	54.09	kJ/mol	Joback Method
log10ws	-4.69		Crippen Method
logp	4.156		Crippen Method
mcvol	179.550	ml/mol	McGowan Method
pc	2495.01	kPa	Joback Method
rinpola	1632.00		NIST Webbook
tb	594.20	K	Joback Method
tc	806.18	K	Joback Method
tf	345.97	K	Joback Method
vc	0.679	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	416.58	J/molxK	594.20	Joback Method
cpg	431.71	J/molxK	629.53	Joback Method
cpg	445.98	J/molxK	664.86	Joback Method
cpg	459.43	J/molxK	700.19	Joback Method
cpg	472.08	J/molxK	735.52	Joback Method
cpg	483.96	J/molxK	770.85	Joback Method
cpg	495.11	J/molxK	806.18	Joback Method
dvisc	0.0015238	Paxs	345.97	Joback Method
dvisc	0.0008663	Paxs	387.34	Joback Method

dvisc	0.0005492	Paxs	428.71	Joback Method
dvisc	0.0003772	Paxs	470.09	Joback Method
dvisc	0.0002754	Paxs	511.46	Joback Method
dvisc	0.0002107	Paxs	552.83	Joback Method
dvisc	0.0001673	Paxs	594.20	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U378188&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U378188&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/61-889-8/2-Bromobenzyl-alcohol-n-pentyl-ether.pdf>

Generated by Cheméo on 2024-04-24 02:16:26.512869443 +0000 UTC m=+16214235.433446754.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.