

# Chlorphenesin carbamate

**Other names:**

1,2-Propanediol, 3-(4-chlorophenoxy)-, 1-carbamate  
1,2-Propanediol, 3-(p-chlorophenoxy)-, 1-carbamate  
Maolate  
U-19,646  
3-(p-Chlorophenoxy)-2-Hydroxypropyl carbamate  
3-(4-Chlorophenoxy)-2-hydroxypropyl carbamate  
Carbamic acid, 3-(p-chlorophenoxy)-2-hydroxypropyl ester  
Rinlaxer  
U 19646  
3-(p-Chlorophenoxy)-1,2-propanediol 1-carbamate  
3-(4-Chlorophenoxy)-1,2-propanediol-1-carbamate  
NSC 82943  
Chlorphenesin

**Inchi:**

InChI=1S/C10H12ClNO4/c11-7-1-3-9(4-2-7)15-5-8(13)6-16-10(12)14/h1-4,8,13H,5-6H2,(

**InchiKey:**

SKPLBLUECSEIFO-UHFFFAOYSA-N

**Formula:**

C10H12ClNO4

**SMILES:**

NC(=O)OCC(O)COc1ccc(Cl)cc1

**Mol. weight [g/mol]:**

245.66

**CAS:**

886-74-8

## Physical Properties

Property code	Value	Unit	Source
gf	-287.56	kJ/mol	Joback Method
hf	-541.15	kJ/mol	Joback Method
hfus	29.24	kJ/mol	Joback Method
hvap	83.67	kJ/mol	Joback Method
log10ws	-2.18		Crippen Method
logp	1.175		Crippen Method
mccvol	169.400	ml/mol	McGowan Method
pc	3407.89	kPa	Joback Method
rinpol	1695.00		NIST Webbook
rinpol	1677.00		NIST Webbook
tb	760.27	K	Joback Method
tc	972.14	K	Joback Method
tf	494.79	K	Joback Method
vc	0.621	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	451.89	J/mol×K	760.27	Joback Method
cpg	461.55	J/mol×K	795.58	Joback Method
cpg	470.47	J/mol×K	830.89	Joback Method
cpg	478.66	J/mol×K	866.20	Joback Method
cpg	486.12	J/mol×K	901.51	Joback Method
cpg	492.87	J/mol×K	936.83	Joback Method
cpg	498.89	J/mol×K	972.14	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C886748&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C886748&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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