

# 2,2-Dimethylpropanoic acid, 3,4-dichlorophenyl ester

Inchi:	InChI=1S/C11H12Cl2O2/c1-11(2,3)10(14)15-7-4-5-8(12)9(13)6-7/h4-6H,1-3H3
InchiKey:	XIRSVOKAKYHVVOX-UHFFFAOYSA-N
Formula:	C11H12Cl2O2
SMILES:	CC(C)(C)C(=O)Oc1ccc(Cl)c(Cl)c1
Mol. weight [g/mol]:	247.12

## Physical Properties

Property code	Value	Unit	Source
gf	-120.05	kJ/mol	Joback Method
hf	-341.81	kJ/mol	Joback Method
hfus	21.28	kJ/mol	Joback Method
hvap	60.31	kJ/mol	Joback Method
log10ws	-4.17		Crippen Method
logp	3.945		Crippen Method
mcvol	174.010	ml/mol	McGowan Method
pc	2545.61	kPa	Joback Method
rinpol	1552.00		NIST Webbook
rinpol	1552.00		NIST Webbook
tb	635.64	K	Joback Method
tc	867.66	K	Joback Method
tf	399.61	K	Joback Method
vc	0.654	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	400.40	J/molxK	635.64	Joback Method
cpg	413.12	J/molxK	674.31	Joback Method
cpg	424.92	J/molxK	712.98	Joback Method
cpg	435.84	J/molxK	751.65	Joback Method
cpg	445.91	J/molxK	790.32	Joback Method
cpg	455.19	J/molxK	828.99	Joback Method
cpg	463.72	J/molxK	867.66	Joback Method
dvisc	0.0012318	Paxs	399.61	Joback Method

dvisc	0.0007473	Paxs	438.95	Joback Method
dvisc	0.0004922	Paxs	478.29	Joback Method
dvisc	0.0003454	Paxs	517.62	Joback Method
dvisc	0.0002548	Paxs	556.96	Joback Method
dvisc	0.0001957	Paxs	596.30	Joback Method
dvisc	0.0001553	Paxs	635.64	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U308044&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U308044&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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