

# Cyclopropanecarboxylic acid, 2,2-dichloro-1-methyl-, methyl ester

<b>Other names:</b>	Methyl 2,2-dichloro-1-methylcyclopropanecarboxylate
<b>Inchi:</b>	InChI=1S/C6H8Cl2O2/c1-5(4(9)10-2)3-6(5,7)8/h3H2,1-2H3
<b>InchiKey:</b>	JGCAZFUFORGMFB-UHFFFAOYSA-N
<b>Formula:</b>	C6H8Cl2O2
<b>SMILES:</b>	<chem>COC(=O)C1(C)CC1(Cl)Cl</chem>
<b>Mol. weight [g/mol]:</b>	183.03
<b>CAS:</b>	1447-13-8

## Physical Properties

Property code	Value	Unit	Source
gf	-216.08	kJ/mol	Joback Method
hf	-360.51	kJ/mol	Joback Method
hfus	9.09	kJ/mol	Joback Method
hvap	44.18	kJ/mol	Joback Method
log10ws	-1.76		Crippen Method
logp	1.743		Crippen Method
mcvol	116.460	ml/mol	McGowan Method
pc	3740.80	kPa	Joback Method
tb	490.38	K	Joback Method
tc	713.66	K	Joback Method
tf	350.88	K	Joback Method
vc	0.446	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	235.61	J/mol×K	490.38	Joback Method
cpg	245.46	J/mol×K	527.59	Joback Method
cpg	254.30	J/mol×K	564.81	Joback Method
cpg	262.34	J/mol×K	602.02	Joback Method
cpg	269.78	J/mol×K	639.23	Joback Method
cpg	276.85	J/mol×K	676.45	Joback Method
cpg	283.75	J/mol×K	713.66	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	347.20	K	1.00	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1447138&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1447138&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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