

Benzene, 1,3-dibromo-

Other names:	1,3-DIBROMOBENZENE Benzene, m-dibromo- m-Dibromobenzene
Inchi:	InChI=1S/C6H4Br2/c7-5-2-1-3-6(8)4-5/h1-4H
InchiKey:	JSRLURSZEMPLAFO-UHFFFAOYSA-N
Formula:	C6H4Br2
SMILES:	Brc1cccc(Br)c1
Mol. weight [g/mol]:	235.90
CAS:	108-36-1

Physical Properties

Property code	Value	Unit	Source
gf	131.06	kJ/mol	Joback Method
hf	110.55	kJ/mol	Joback Method
hfp _i	1000.00 ± 10.00	kJ/mol	NIST Webbook
hfp _{iz}	1020.00 ± 10.00	kJ/mol	NIST Webbook
hfus	15.52	kJ/mol	Joback Method
h _{vap}	44.76	kJ/mol	Joback Method
ie	9.01 ± 0.02	eV	NIST Webbook
ie	8.95 ± 0.02	eV	NIST Webbook
ie	9.10	eV	NIST Webbook
ie	9.05 ± 0.03	eV	NIST Webbook
ie	9.07	eV	NIST Webbook
ie	8.85	eV	NIST Webbook
ie	9.01 ± 0.05	eV	NIST Webbook
log ₁₀ ws	-3.54		Aqueous Solubility Prediction Method
log ₁₀ ws	-3.54		Estimated Solubility Method
logp	3.212		Crippen Method
m _{cvol}	106.640	ml/mol	McGowan Method
pc	5569.17	kPa	Joback Method
rin _{pol}	1190.00		NIST Webbook
rin _{pol}	1197.00		NIST Webbook
rin _{pol}	1197.00		NIST Webbook
rin _{pol}	1197.00		NIST Webbook
rin _{pol}	1162.20		NIST Webbook
rin _{pol}	1190.00		NIST Webbook

rinpol	1162.20		NIST Webbook
ripol	1756.00		NIST Webbook
ripol	1756.00		NIST Webbook
tb	491.20	K	NIST Webbook
tc	757.23	K	Joback Method
tf	266.25 ± 0.50	K	NIST Webbook
tf	266.00 ± 1.50	K	NIST Webbook
vc	0.388	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	202.93	J/mol×K	757.23	Joback Method
cpg	163.98	J/mol×K	500.66	Joback Method
cpg	172.09	J/mol×K	543.42	Joback Method
cpg	179.47	J/mol×K	586.18	Joback Method
cpg	186.19	J/mol×K	628.94	Joback Method
cpg	192.30	J/mol×K	671.70	Joback Method
cpg	197.86	J/mol×K	714.47	Joback Method
cpl	178.08	J/mol×K	295.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	176.01	J/mol×K	283.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	176.40	J/mol×K	285.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	176.71	J/mol×K	287.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes

cpl	177.04	J/mol×K	289.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes	
cpl	177.38	J/mol×K	291.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes	
cpl	177.72	J/mol×K	293.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes	
cpl	189.09	J/mol×K	349.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes	
cpl	178.44	J/mol×K	297.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes	
cpl	178.81	J/mol×K	299.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes	
cpl	179.19	J/mol×K	301.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes	
cpl	179.58	J/mol×K	303.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes	
cpl	179.97	J/mol×K	305.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes	

cpl	180.37	J/mol×K	307.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	180.78	J/mol×K	309.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	181.19	J/mol×K	311.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	181.60	J/mol×K	313.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	182.01	J/mol×K	315.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	182.43	J/mol×K	317.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	182.85	J/mol×K	319.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	183.28	J/mol×K	321.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes

cpl	189.47	J/mol×K	351.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	184.13	J/mol×K	325.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	184.55	J/mol×K	327.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	184.98	J/mol×K	329.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	185.40	J/mol×K	331.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	185.83	J/mol×K	333.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	186.25	J/mol×K	335.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	186.67	J/mol×K	337.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	187.08	J/mol×K	339.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes

cpl	187.49	J/mol×K	341.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	187.90	J/mol×K	343.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	188.30	J/mol×K	345.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	188.70	J/mol×K	347.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	192.00	J/mol×K	298.15	NIST Webbook
cpl	183.70	J/mol×K	323.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
cpl	189.85	J/mol×K	353.15	Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Bromochloro-Substituted Benzenes
dvisc	0.0004259	Pa×s	469.87	Joback Method
dvisc	0.0018711	Pa×s	315.92	Joback Method
dvisc	0.0012528	Pa×s	346.71	Joback Method
dvisc	0.0008955	Pa×s	377.50	Joback Method
dvisc	0.0005270	Pa×s	439.08	Joback Method
dvisc	0.0003533	Pa×s	500.66	Joback Method
dvisc	0.0006734	Pa×s	408.29	Joback Method
hfust	14.22	kJ/mol	266.25	NIST Webbook
hfust	13.21	kJ/mol	266.30	NIST Webbook
hfust	13.21	kJ/mol	266.30	NIST Webbook
hvapt	48.30	kJ/mol	458.50	NIST Webbook
sfust	53.40	J/mol×K	266.25	NIST Webbook

Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	339.20	K	0.70	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.43239e+01
Coeff. B	-4.02376e+03
Coeff. C	-7.66160e+01
Temperature range (K), min.	363.28
Temperature range (K), max.	523.09

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	9.02731e+01
Coeff. B	-9.55542e+03
Coeff. C	-1.08844e+01
Coeff. D	5.13030e-06
Temperature range (K), min.	266.25
Temperature range (K), max.	761.00

Sources

KDB: <https://www.thermo.com/files/research/kdb/mol/mol1675.mol>

Estimated Solubility Method: http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt

Heat Capacities and Densities of Some Liquid Chloro-, Bromo-, and Iodo-substituted Benzenes: <https://www.doi.org/10.1021/jc600573w>

McGowan, M. S. <http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook: <http://webbook.nist.gov/cgi/cbook.cgi?ID=C108361&Units=SI>

Aqueous Solubility Prediction Method: <http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx>

The Yaws Handbook of Vapor Pressure: <https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

KDB Vapor Pressure Data: <https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=1675>
Crippen Method: <http://pubs.acs.org/doi/abs/10.1021/ci990307l>
Joback Method: https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
cpl:	Liquid phase heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfpi:	Enthalpy of formation of positive ion at standard conditions
hfpiz:	Enthalpy of formation of positive ion at 0K
hfus:	Enthalpy of fusion at standard conditions
hfust:	Enthalpy of fusion at a given temperature
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
rinpol:	Non-polar retention indices
ripol:	Polar retention indices
sfust:	Entropy of fusion at a given temperature
tb:	Normal Boiling Point Temperature
tbrp:	Boiling point at reduced pressure
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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