

# Carbonotrithioic acid, dimethyl ester

<b>Other names:</b>	Carbonic acid, trithio-, dimethyl ester Dimethyl trithiocarbonate (CH <sub>3</sub> S) <sub>2</sub> CS Dimethylester kyseliny trithiuhlicite Trithiocarbonic acid dimethyl ester
<b>Inchi:</b>	InChI=1S/C3H6S3/c1-5-3(4)6-2/h1-2H3
<b>InchiKey:</b>	IQWMXKTYXNMSLC-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>3</sub> H <sub>6</sub> S <sub>3</sub>
<b>SMILES:</b>	CSC(=S)SC
<b>Mol. weight [g/mol]:</b>	138.28
<b>CAS:</b>	2314-48-9

## Physical Properties

Property code	Value	Unit	Source
chl	-3836.00 ± 4.20	kJ/mol	NIST Webbook
gf	157.68	kJ/mol	Joback Method
hf	57.70	kJ/mol	NIST Webbook
hfl	-8.40 ± 4.20	kJ/mol	NIST Webbook
hfus	16.39	kJ/mol	Joback Method
hvap	66.27	kJ/mol	NIST Webbook
hvap	66.10	kJ/mol	NIST Webbook
ie	8.50	eV	NIST Webbook
ie	7.90	eV	NIST Webbook
log10ws	-2.19		Crippen Method
logp	1.997		Crippen Method
mvol	97.880	ml/mol	McGowan Method
pc	5251.00	kPa	Joback Method
rinpol	1195.80		NIST Webbook
rinpol	1195.80		NIST Webbook
tb	475.64	K	Joback Method
tc	730.36	K	Joback Method
tf	266.90 ± 0.60	K	NIST Webbook
vc	0.347	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	161.88	J/mol×K	475.64	Joback Method
cpg	169.20	J/mol×K	518.09	Joback Method
cpg	175.97	J/mol×K	560.55	Joback Method
cpg	182.23	J/mol×K	603.00	Joback Method
cpg	188.01	J/mol×K	645.45	Joback Method
cpg	193.34	J/mol×K	687.91	Joback Method
cpg	198.26	J/mol×K	730.36	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2314489&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2314489&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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