

# Phorate

<b>Other names:</b>	AC 3911 Agrimet American Cyanamid 3,911 Dithiophosphate de O,O-diethyle et D'ethylthiomethyle EI3911 ENT 24,042 Experimental Insecticide 3911 Foraat Forate Geomet Granutox L 11/6 Methanethiol, (ethylthio)-, S-ester with O,O-diethyl phosphorodithioic acid O,O-Diaethyl-S-(aethylthio-methyl)-dithiophosphat O,O-Diethyl S-Ethylmercaptomethyl dithiophosphate O,O-Diethyl S-[(ethylthio)methyl] phosphorodithioate O,O-Diethyl S-ethylthiomethyl dithiophosphate O,O-Diethyl S-ethylthiomethyl dithiophosphonate O,O-Diethyl S-ethylthiomethyl thiothionophosphate O,O-Diethyl ethylthiomethyl phosphorodithioate O,O-Diethyl-S-(ethylthio-methyl)-dithiofosfaat O,O-Dietil-S-(etiltio-metil)-ditiofosfato Phorat Phorate 10G Phosphorodithioic acid, O,O-diethyl S-[(ethylthio)methyl] ester Rampart Rcra waste number P094 Thimate Thimenox Thimet Thimet 10G Thimet G Timet VUAgT 182 Vegfru Vegfru foratox Volphor
<b>Inchi:</b>	InChI=1S/C7H17O2PS3/c1-4-8-10(11,9-5-2)13-7-12-6-3/h4-7H2,1-3H3
<b>InchiKey:</b>	BULVZWIRKLYCBC-UHFFFAOYSA-N
<b>Formula:</b>	C7H17O2PS3

**SMILES:** CCOP(=S)(OCC)SCSCC  
**Mol. weight [g/mol]:** 260.38  
**CAS:** 298-02-2

## Physical Properties

Property code	Value	Unit	Source
hvap	70.80	kJ/mol	NIST Webbook
log10ws	-4.11		Aqueous Solubility Prediction Method
log10ws	-4.11		Estimated Solubility Method
logp	3.728		Crippen Method
mcvol	190.740	ml/mol	McGowan Method
rinpol	1703.00		NIST Webbook
rinpol	1705.00		NIST Webbook
rinpol	1675.00		NIST Webbook
rinpol	1686.00		NIST Webbook
rinpol	1691.00		NIST Webbook
rinpol	1688.00		NIST Webbook
rinpol	1700.00		NIST Webbook
rinpol	1641.00		NIST Webbook
rinpol	1675.00		NIST Webbook
rinpol	1688.00		NIST Webbook
rinpol	1675.00		NIST Webbook
rinpol	1703.00		NIST Webbook
rinpol	1686.00		NIST Webbook
ripol	2335.00		NIST Webbook
ripol	2335.00		NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.99169e+01
Coeff. B	-8.51266e+03
Temperature range (K), min.	433.67
Temperature range (K), max.	582.84

---

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C298022&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C298022&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure: Crippen Method:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a> <a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa</a>
<b>Estimated Solubility Method:</b>	<a href="http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt">http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt</a>

## Legend

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>vap</sub>:</b>	Vapor pressure
<b>ri<sub>npol</sub>:</b>	Non-polar retention indices
<b>ri<sub>pol</sub>:</b>	Polar retention indices

Latest version available from:

<https://www.chemeo.com/cid/62-122-7/Phorate.pdf>

Generated by Cheméo on 2024-05-01 05:40:07.78608434 +0000 UTC m=+16831256.706661652.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.