

# 1,2-Ethanediol, 1-phenyl-

<b>Other names:</b>	Styrene glycol «alpha», «beta»-Dihydroxyethylbenzene Phenyl glycol Phenylethanediol Phenylethylene glycol Styryl alcohol 1-Phenyl-1,2-ethanediol 1,2-Dihydroxy-1-phenylethane Phenyl-1,2-ethanediol 1,2-Dihydroxyethylbenzene 1,2-Ethanediol, phenyl- 1-Fenyl-1,2-ethandiol Fenylglycol 1-Phenylethylene glycol NSC 406601 phenylethane-1,2-diol
<b>Inchi:</b>	InChI=1S/C8H10O2/c9-6-8(10)7-4-2-1-3-5-7/h1-5,8-10H,6H2
<b>InchiKey:</b>	PWMWNFMRSKOCEY-UHFFFAOYSA-N
<b>Formula:</b>	C8H10O2
<b>SMILES:</b>	OCC(O)c1ccccc1
<b>Mol. weight [g/mol]:</b>	138.16
<b>CAS:</b>	93-56-1

## Physical Properties

Property code	Value	Unit	Source
gf	-147.19	kJ/mol	Joback Method
hf	-281.66	kJ/mol	Joback Method
hfus	15.17	kJ/mol	Joback Method
hvap	68.65	kJ/mol	Joback Method
log10ws	-1.26		Crippen Method
logp	0.712		Crippen Method
mcvol	111.560	ml/mol	McGowan Method
pc	4672.09	kPa	Joback Method
rinpol	1384.00		NIST Webbook
rinpol	1308.10		NIST Webbook
rinpol	1372.00		NIST Webbook
rinpol	1384.00		NIST Webbook

ripol	1308.10		NIST Webbook
ripol	1372.00		NIST Webbook
ripol	1735.00		NIST Webbook
ripol	1735.00		NIST Webbook
tb	546.20	K	NIST Webbook
tc	782.98	K	Joback Method
tf	341.20 ± 2.00	K	NIST Webbook
tf	338.65 ± 1.00	K	NIST Webbook
tf	340.15 ± 2.00	K	NIST Webbook
tf	336.00 ± 4.00	K	NIST Webbook
vc	0.407	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	269.26	J/mol×K	593.04	Joback Method
cpg	278.28	J/mol×K	624.70	Joback Method
cpg	286.77	J/mol×K	656.35	Joback Method
cpg	294.75	J/mol×K	688.01	Joback Method
cpg	302.25	J/mol×K	719.66	Joback Method
cpg	309.29	J/mol×K	751.32	Joback Method
cpg	315.89	J/mol×K	782.98	Joback Method
dvisc	0.0339541	Paxs	312.98	Joback Method
dvisc	0.0050951	Paxs	359.66	Joback Method
dvisc	0.0011821	Paxs	406.33	Joback Method
dvisc	0.0003706	Paxs	453.01	Joback Method
dvisc	0.0001443	Paxs	499.69	Joback Method
dvisc	0.0000660	Paxs	546.36	Joback Method
dvisc	0.0000342	Paxs	593.04	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C93561&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C93561&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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