

# Diethylmalonic acid, monochloride, 1-phenylpropyl ester

<b>Inchi:</b>	InChI=1S/C16H21ClO3/c1-4-13(12-10-8-7-9-11-12)20-15(19)16(5-2,6-3)14(17)18/h7-11,
<b>InchiKey:</b>	ZXWZQVWUEBQBKK-UHFFFAOYSA-N
<b>Formula:</b>	C16H21ClO3
<b>SMILES:</b>	CCC(OC(=O)C(CC)(CC)C(=O)Cl)c1ccccc1
<b>Mol. weight [g/mol]:</b>	296.79

## Physical Properties

Property code	Value	Unit	Source
gf	-178.12	kJ/mol	Joback Method
hf	-524.19	kJ/mol	Joback Method
hfus	28.88	kJ/mol	Joback Method
hvap	72.09	kJ/mol	Joback Method
log10ws	-4.63		Crippen Method
logp	4.253		Crippen Method
mcvol	233.790	ml/mol	McGowan Method
pc	1851.52	kPa	Joback Method
rinpol	1813.00		NIST Webbook
rinpol	1813.00		NIST Webbook
tb	756.08	K	Joback Method
tc	974.23	K	Joback Method
tf	435.93	K	Joback Method
vc	0.885	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	651.55	J/molxK	756.08	Joback Method
cpg	716.29	J/molxK	937.87	Joback Method
cpg	705.32	J/molxK	901.51	Joback Method
cpg	693.42	J/molxK	865.15	Joback Method
cpg	680.54	J/molxK	828.80	Joback Method
cpg	666.60	J/molxK	792.44	Joback Method
cpg	726.40	J/molxK	974.23	Joback Method
dvisc	0.0000817	Paxs	756.08	Joback Method

dvisc	0.0001095	Paxs	702.72	Joback Method
dvisc	0.0001540	Paxs	649.36	Joback Method
dvisc	0.0002301	Paxs	596.01	Joback Method
dvisc	0.0003723	Paxs	542.65	Joback Method
dvisc	0.0006687	Paxs	489.29	Joback Method
dvisc	0.0013863	Paxs	435.93	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370193&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370193&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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