

# Benzeneacetic acid, 3,4-dimethoxy-

<b>Other names:</b>	Acetic acid, (3,4-dimethoxyphenyl)- (3,4-Dimethoxyphenyl)acetic acid Homoveratric Acid 3,4-(Dimethoxy)benzeneacetic acid
<b>Inchi:</b>	InChI=1S/C10H12O4/c1-13-8-4-3-7(6-10(11)12)5-9(8)14-2/h3-5H,6H2,1-2H3,(H,11,12)
<b>InchiKey:</b>	WUAXWQRULBZETB-UHFFFAOYSA-N
<b>Formula:</b>	C10H12O4
<b>SMILES:</b>	COc1ccc(CC(=O)O)cc1OC
<b>Mol. weight [g/mol]:</b>	196.20
<b>CAS:</b>	93-40-3

## Physical Properties

Property code	Value	Unit	Source
gf	-349.27	kJ/mol	Joback Method
hf	-565.39	kJ/mol	Joback Method
hfus	22.98	kJ/mol	Joback Method
hvap	69.70	kJ/mol	Joback Method
log10ws	-1.61		Crippen Method
logp	1.331		Crippen Method
mcvol	147.180	ml/mol	McGowan Method
pc	3243.03	kPa	Joback Method
tb	655.73	K	Joback Method
tc	854.45	K	Joback Method
tf	409.13	K	Joback Method
vc	0.548	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	372.72	J/molxK	655.73	Joback Method
cpg	383.42	J/molxK	688.85	Joback Method
cpg	393.54	J/molxK	721.97	Joback Method
cpg	403.09	J/molxK	755.09	Joback Method
cpg	412.05	J/molxK	788.21	Joback Method

cpg	420.44	J/mol×K	821.33	Joback Method
cpg	428.23	J/mol×K	854.45	Joback Method
dvisc	0.0010962	Paxs	409.13	Joback Method
dvisc	0.0005142	Paxs	450.23	Joback Method
dvisc	0.0002737	Paxs	491.33	Joback Method
dvisc	0.0001606	Paxs	532.43	Joback Method
dvisc	0.0001017	Paxs	573.53	Joback Method
dvisc	0.0000685	Paxs	614.63	Joback Method
dvisc	0.0000485	Paxs	655.73	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C93403&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C93403&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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