

# 2-Acetyl-1-phenylhydrazine

<b>Other names:</b>	«beta»-Acetylphenylhydrazine Acetylphenylhydrazine APH Hydracetin N-Acetyl-N'-Phenylhydrazine N'-Phenylacethydrazide Pyrodin Pyrodine 1-Acetyl-2-phenylhydrazine 2-Phenylacetohydrazide Acetic acid, 2-phenylhydrazide Acetic acid phenylhydrazone 1-Phenyl-2-acetylhydrazine Fenylhydrazid kyseliny octove N-(Phenylamino)acetamide NSC 2064 2'-phenylacetohydrazide
<b>Inchi:</b>	InChI=1S/C8H10N2O/c1-7(11)9-10-8-5-3-2-4-6-8/h2-6,10H,1H3,(H,9,11)
<b>InchiKey:</b>	UICBCXONCUFSOI-UHFFFAOYSA-N
<b>Formula:</b>	C8H10N2O
<b>SMILES:</b>	CC(=O)NNc1ccccc1
<b>Mol. weight [g/mol]:</b>	150.18
<b>CAS:</b>	114-83-0

## Physical Properties

Property code	Value	Unit	Source
gf	178.75	kJ/mol	Joback Method
hf	22.44	kJ/mol	Joback Method
hfus	22.31	kJ/mol	Joback Method
hvap	55.30	kJ/mol	Joback Method
log10ws	-1.85		Crippen Method
logp	1.150		Crippen Method
mcvol	121.350	ml/mol	McGowan Method
pc	4077.71	kPa	Joback Method
tb	563.33	K	Joback Method
tc	785.80	K	Joback Method
tf	361.59	K	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	278.91	J/mol×K	563.33	Joback Method
cpg	291.01	J/mol×K	600.41	Joback Method
cpg	302.28	J/mol×K	637.49	Joback Method
cpg	312.75	J/mol×K	674.57	Joback Method
cpg	322.47	J/mol×K	711.64	Joback Method
cpg	331.46	J/mol×K	748.72	Joback Method
cpg	339.78	J/mol×K	785.80	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C114830&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C114830&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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