

# 2-(Hexyloxy)benzaldehyde

<b>Other names:</b>	o-Hexyloxybenzaldehyde 2-n-Hexyloxybenzaldehyde Benzaldehyde, 2-(hexyloxy)- Benzaldehyde, o-(n-hexyloxy)-
<b>Inchi:</b>	InChI=1S/C13H18O2/c1-2-3-4-7-10-15-13-9-6-5-8-12(13)11-14/h5-6,8-9,11H,2-4,7,10H2
<b>InchiKey:</b>	IFOIDROUJIGQAV-UHFFFAOYSA-N
<b>Formula:</b>	C13H18O2
<b>SMILES:</b>	CCCCCCOc1cccc1C=O
<b>Mol. weight [g/mol]:</b>	206.28
<b>CAS:</b>	7162-59-6

## Physical Properties

Property code	Value	Unit	Source
gf	-43.16	kJ/mol	Joback Method
hf	-304.39	kJ/mol	Joback Method
hfus	26.55	kJ/mol	Joback Method
hvap	56.60	kJ/mol	Joback Method
log10ws	-3.93		Crippen Method
logp	3.458		Crippen Method
mcvol	177.710	ml/mol	McGowan Method
pc	2282.77	kPa	Joback Method
tb	599.58	K	Joback Method
tc	799.07	K	Joback Method
tf	339.44	K	Joback Method
vc	0.691	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	443.39	J/molxK	599.58	Joback Method
cpg	458.64	J/molxK	632.83	Joback Method
cpg	473.07	J/molxK	666.08	Joback Method
cpg	486.73	J/molxK	699.33	Joback Method
cpg	499.61	J/molxK	732.57	Joback Method

cpg	511.75	J/molxK	765.82	Joback Method
cpg	523.16	J/molxK	799.07	Joback Method
dvisc	0.0018840	Paxs	339.44	Joback Method
dvisc	0.0010225	Paxs	382.80	Joback Method
dvisc	0.0006285	Paxs	426.15	Joback Method
dvisc	0.0004226	Paxs	469.51	Joback Method
dvisc	0.0003039	Paxs	512.87	Joback Method
dvisc	0.0002301	Paxs	556.22	Joback Method
dvisc	0.0001813	Paxs	599.58	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	405.50 ± 2.50	K	0.07	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7162596&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7162596&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure

<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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