

rubidium nitrate

Inchi:	InChI=1S/NO3.Rb/c2-1(3)4;/q-1;+1
InchiKey:	RTHYXYOJKHGZJT-UHFFFAOYSA-N
Formula:	NO ₃ Rb
SMILES:	O=[N+](=[O-])[O-].[Rb+]
Mol. weight [g/mol]:	147.47
CAS:	13126-12-0

Physical Properties

Property code	Value	Unit	Source
hfus	3.83	kJ/mol	Determination of the Equilibrium Temperatures and Enthalpies of the Solid-Solid Transitions of Rubidium Nitrate by Differential Scanning Calorimetry
ie	8.89 ± 0.03	eV	NIST Webbook
tf	582.65	K	Crystallization of ionic salts for calibration of differential scanning calorimeters
tt	583.00	K	Heat capacities and enthalpies of fusion of lithium and rubidium nitrates. Heat capacities, enthalpies of fusion and enthalpies of formation of the intermediate compounds Ag0.5Rb0.5NO ₃ and Li0.5Rb0.5NO ₃ .

Sources

Crystallization of ionic salts for calibration of differential scanning NIST Webbook:

<https://www.doi.org/10.1016/j.tca.2016.08.003>

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C13126120&Units=SI>

Determination of the Equilibrium Temperatures and Enthalpies of the Solid-Solid Transitions of Rubidium Nitrate by Differential Scanning Calorimetry, Heat capacities, enthalpies of fusion and enthalpies of formation of the intermediate compounds Ag0.5Rb0.5NO₃ and Li0.5Rb0.5NO₃:

<https://www.doi.org/10.1016/j.tca.2007.12.008>

<https://www.doi.org/10.1016/j.tca.2013.06.040>

Legend

hfus:	Enthalpy of fusion at standard conditions
ie:	Ionization energy
tf:	Normal melting (fusion) point
tt:	Triple Point Temperature

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<https://www.chemeo.com/cid/62-432-3/rubidium-nitrate.pdf>

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