

# Benzene, (1-ethyloctyl)-

<b>Other names:</b>	Decane, 3-phenyl- (3-Decyl)benzene (1-Ethyloctyl)benzene
<b>Inchi:</b>	InChI=1S/C16H26/c1-3-5-6-7-9-12-15(4-2)16-13-10-8-11-14-16/h8,10-11,13-15H,3-7,9,1
<b>InchiKey:</b>	PYVIFMPVFLOTLN-UHFFFAOYSA-N
<b>Formula:</b>	C16H26
<b>SMILES:</b>	CCCCCCCC(CC)c1ccccc1
<b>Mol. weight [g/mol]:</b>	218.38
<b>CAS:</b>	4621-36-7

## Physical Properties

Property code	Value	Unit	Source
gf	193.81	kJ/mol	Joback Method
hf	-142.32	kJ/mol	Joback Method
hfus	27.71	kJ/mol	Joback Method
hvap	53.10	kJ/mol	Joback Method
log10ws	-5.59		Crippen Method
logp	5.541		Crippen Method
mcvol	212.540	ml/mol	McGowan Method
pc	1714.61	kPa	Joback Method
rinpol	1553.00		NIST Webbook
rinpol	267.30		NIST Webbook
rinpol	1573.00		NIST Webbook
rinpol	1553.00		NIST Webbook
rinpol	1560.00		NIST Webbook
ripol	1767.00		NIST Webbook
ripol	1767.00		NIST Webbook
tb	591.72	K	Joback Method
tc	784.61	K	Joback Method
tf	281.50	K	Joback Method
vc	0.818	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	545.09	J/mol×K	591.72	Joback Method
cpg	564.58	J/mol×K	623.87	Joback Method
cpg	583.04	J/mol×K	656.02	Joback Method
cpg	600.51	J/mol×K	688.17	Joback Method
cpg	617.03	J/mol×K	720.31	Joback Method
cpg	632.64	J/mol×K	752.46	Joback Method
cpg	647.38	J/mol×K	784.61	Joback Method
dvisc	0.0047483	Paxs	281.50	Joback Method
dvisc	0.0016769	Paxs	333.20	Joback Method
dvisc	0.0007833	Paxs	384.91	Joback Method
dvisc	0.0004381	Paxs	436.61	Joback Method
dvisc	0.0002772	Paxs	488.31	Joback Method
dvisc	0.0001914	Paxs	540.02	Joback Method
dvisc	0.0001410	Paxs	591.72	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4621367&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4621367&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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