

# D-Ribonitrile, 2,3,4,5-tetraacetate

<b>Other names:</b>	Ribonitrile, 2,3,4,5-tetraacetate, D-D-(-)-Ribose, aldonitrile, tetraacetate
<b>Inchi:</b>	InChI=1S/C13H17NO8/c1-7(15)19-6-12(21-9(3)17)13(22-10(4)18)11(5-14)20-8(2)16/h11
<b>InchiKey:</b>	YHTPKBYAZJOQCI-UHFFFAOYSA-N
<b>Formula:</b>	C13H17NO8
<b>SMILES:</b>	CC(=O)OCC(OC(C)=O)C(OC(C)=O)C(C#N)OC(C)=O
<b>Mol. weight [g/mol]:</b>	315.28
<b>CAS:</b>	25546-50-3

## Physical Properties

Property code	Value	Unit	Source
gf	-751.24	kJ/mol	Joback Method
hf	-1141.81	kJ/mol	Joback Method
hfus	31.51	kJ/mol	Joback Method
hvap	90.47	kJ/mol	Joback Method
log10ws	-0.92		Crippen Method
logp	-0.132		Crippen Method
mcvol	225.170	ml/mol	McGowan Method
pc	1918.62	kPa	Joback Method
rinpol	1691.10		NIST Webbook
rinpol	1691.10		NIST Webbook
tb	902.76	K	Joback Method
tc	1116.10	K	Joback Method
tf	544.90	K	Joback Method
vc	0.868	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	675.92	J/molxK	902.76	Joback Method
cpg	685.12	J/molxK	938.32	Joback Method
cpg	693.11	J/molxK	973.87	Joback Method
cpg	699.85	J/molxK	1009.43	Joback Method
cpg	705.31	J/molxK	1044.99	Joback Method

cpg	709.46	J/mol×K	1080.55	Joback Method
cpg	712.26	J/mol×K	1116.10	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C25546503&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C25546503&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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