

# 4-Tetradecanone

<b>Other names:</b>	tetradecan-4-one
<b>Inchi:</b>	InChI=1S/C14H28O/c1-3-5-6-7-8-9-10-11-13-14(15)12-4-2/h3-13H2,1-2H3
<b>InchiKey:</b>	OWIOJZLDFONFSU-UHFFFAOYSA-N
<b>Formula:</b>	C14H28O
<b>SMILES:</b>	CCCCCCCCCCC(=O)CCC
<b>Mol. weight [g/mol]:</b>	212.37
<b>CAS:</b>	26496-20-8

## Physical Properties

Property code	Value	Unit	Source
gf	-61.92	kJ/mol	Joback Method
hf	-444.87	kJ/mol	Joback Method
hfus	33.61	kJ/mol	Joback Method
hvap	53.50	kJ/mol	Joback Method
log10ws	-4.96		Crippen Method
logp	4.886		Crippen Method
mcvol	209.690	ml/mol	McGowan Method
pc	1601.28	kPa	Joback Method
rhoc	235.73 ± 6.37	kg/m <sup>3</sup>	NIST Webbook
tb	573.59	K	Joback Method
tc	725.20 ± 3.70	K	NIST Webbook
tf	297.47	K	Joback Method
vc	0.826	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	537.57	J/mol×K	573.59	Joback Method
cpg	554.73	J/mol×K	601.60	Joback Method
cpg	571.17	J/mol×K	629.62	Joback Method
cpg	586.91	J/mol×K	657.63	Joback Method
cpg	601.98	J/mol×K	685.64	Joback Method
cpg	616.40	J/mol×K	713.66	Joback Method
cpg	630.18	J/mol×K	741.67	Joback Method

dvisc	0.0040142	Paxs	297.47	Joback Method
dvisc	0.0017253	Paxs	343.49	Joback Method
dvisc	0.0009053	Paxs	389.51	Joback Method
dvisc	0.0005444	Paxs	435.53	Joback Method
dvisc	0.0003608	Paxs	481.55	Joback Method
dvisc	0.0002569	Paxs	527.57	Joback Method
dvisc	0.0001932	Paxs	573.59	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.52122e+01
Coeff. B	-4.83443e+03
Coeff. C	-9.40360e+01
Temperature range (K), min.	417.96
Temperature range (K), max.	582.33

## Sources

<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C26496208&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C26496208&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rhoc:</b>	Critical density
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/62-619-6/4-Tetradecanone.pdf>

Generated by Cheméo on 2024-04-23 20:09:31.161477067 +0000 UTC m=+16192220.082054378.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.