

1-Pyrroline, 2-butanoyl

Inchi:	InChI=1S/C8H13NO/c1-2-4-8(10)7-5-3-6-9-7/h2-6H2,1H3
InchiKey:	HKVIHNPWRLQVTI-UHFFFAOYSA-N
Formula:	C8H13NO
SMILES:	CCCC(=O)C1=NCCC1
Mol. weight [g/mol]:	139.19

Physical Properties

Property code	Value	Unit	Source
gf	68.93	kJ/mol	Joback Method
hf	-122.93	kJ/mol	Joback Method
hfus	16.91	kJ/mol	Joback Method
hvap	47.88	kJ/mol	Joback Method
log10ws	-1.51		Crippen Method
logp	1.590		Crippen Method
mcvol	119.970	ml/mol	McGowan Method
pc	3509.58	kPa	Joback Method
rinpol	1118.00		NIST Webbook
tb	514.10	K	Joback Method
tc	733.02	K	Joback Method
tf	329.81	K	Joback Method
vc	0.467	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	280.15	J/mol×K	514.10	Joback Method
cpg	295.33	J/mol×K	550.59	Joback Method
cpg	309.66	J/mol×K	587.07	Joback Method
cpg	323.16	J/mol×K	623.56	Joback Method
cpg	335.84	J/mol×K	660.05	Joback Method
cpg	347.73	J/mol×K	696.54	Joback Method
cpg	358.83	J/mol×K	733.02	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R66219&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvac:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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