

# Acetoacetic acid isoamyl ester

<b>Other names:</b>	Butanoic acid, 3-oxo-, 3-methylbutyl ester Isoamyl acetoacetate 3-Methylbutyl 3-oxobutanoate Isoamyl acetylacetate Isoamyl «beta»-ketobutyrate Isopentyl «beta»-ketobutyrate isopentyl acetoacetate
<b>Inchi:</b>	InChI=1S/C9H16O3/c1-7(2)4-5-12-9(11)6-8(3)10/h7H,4-6H2,1-3H3
<b>InchiKey:</b>	XHRGPLDMNNGHCX-UHFFFAOYSA-N
<b>Formula:</b>	C9H16O3
<b>SMILES:</b>	CC(=O)CC(=O)OCCC(C)C
<b>Mol. weight [g/mol]:</b>	172.22
<b>CAS:</b>	2308-18-1

## Physical Properties

Property code	Value	Unit	Source
gf	-340.38	kJ/mol	Joback Method
hf	-591.75	kJ/mol	Joback Method
hfus	19.93	kJ/mol	Joback Method
hvap	51.14	kJ/mol	Joback Method
log10ws	-1.49		Crippen Method
logp	1.555		Crippen Method
mcvol	146.680	ml/mol	McGowan Method
pc	2589.85	kPa	Joback Method
tb	491.00	K	NIST Webbook
tc	721.50	K	Joback Method
tf	298.28	K	Joback Method
vc	0.564	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	343.82	J/mol×K	535.04	Joback Method
cpg	356.65	J/mol×K	566.12	Joback Method

cpg	368.94	J/molxK	597.19	Joback Method
cpg	380.69	J/molxK	628.27	Joback Method
cpg	391.90	J/molxK	659.35	Joback Method
cpg	402.58	J/molxK	690.42	Joback Method
cpg	412.73	J/molxK	721.50	Joback Method
dvisc	0.0035617	Paxs	298.28	Joback Method
dvisc	0.0017528	Paxs	337.74	Joback Method
dvisc	0.0010005	Paxs	377.20	Joback Method
dvisc	0.0006351	Paxs	416.66	Joback Method
dvisc	0.0004361	Paxs	456.12	Joback Method
dvisc	0.0003180	Paxs	495.58	Joback Method
dvisc	0.0002429	Paxs	535.04	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2308181&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2308181&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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