

# Benzoic acid, 2-(methylamino)-

|                             |  |
|-----------------------------|--|
| <b>Other names:</b>         | 2-(Methylamino)benzoic acid<br>Anthranilic acid, N-methyl-<br>Kyselina 2-methylaminobenzoova<br>Kyselina N-methylantranilova<br>N-Methyl-2-aminobenzoic acid<br>N-Methyl-o-aminobenzoic acid<br>N-Methylantranilic acid<br>o-(Methylamino)benzoic acid |
| <b>Inchi:</b>               | InChI=1S/C8H9NO2/c1-9-7-5-3-2-4-6(7)8(10)11/h2-5,9H,1H3,(H,10,11)  |
| <b>InchiKey:</b>            | WVMBPWMAQDVZCM-UHFFFAOYSA-N  |
| <b>Formula:</b>             | C8H9NO2  |
| <b>SMILES:</b>              | CNc1cccc1C(=O)O  |
| <b>Mol. weight [g/mol]:</b> | 151.16   |
| <b>CAS:</b>                 | 119-68-6   |

## Physical Properties

| Property code | Value   | Unit    | Source                               |
|---------------|---------|---------|--------------------------------------|
| gf            | -57.09  | kJ/mol  | Joback Method                        |
| hf            | -194.73 | kJ/mol  | Joback Method                        |
| hfus          | 20.91   | kJ/mol  | Joback Method                        |
| hvap          | 66.20   | kJ/mol  | Joback Method                        |
| log10ws       | -2.88   |         | Aqueous Solubility Prediction Method |
| logp          | 1.426   |         | Crippen Method                       |
| mcvol         | 117.240 | ml/mol  | McGowan Method                       |
| pc            | 4432.62 | kPa     | Joback Method                        |
| tb            | 610.32  | K       | Joback Method                        |
| tc            | 817.72  | K       | Joback Method                        |
| tf            | 453.65  | K       | Aqueous Solubility Prediction Method |
| vc            | 0.435   | m3/kmol | Joback Method                        |

## Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-------|------|-----------------|--------|
|---------------|-------|------|-----------------|--------|

|     |        |         |        |               |
|-----|--------|---------|--------|---------------|
| cpg | 278.76 | J/mol×K | 610.32 | Joback Method |
| cpg | 288.23 | J/mol×K | 644.89 | Joback Method |
| cpg | 297.10 | J/mol×K | 679.45 | Joback Method |
| cpg | 305.39 | J/mol×K | 714.02 | Joback Method |
| cpg | 313.13 | J/mol×K | 748.59 | Joback Method |
| cpg | 320.34 | J/mol×K | 783.15 | Joback Method |
| cpg | 327.05 | J/mol×K | 817.72 | Joback Method |

## Sources

**Aqueous Solubility Prediction Method:** <http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx>

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C119686&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

|                 |   |
|-----------------|---|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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