

# 4-Pyridinecarbonitrile

<b>Other names:</b>	.gamma.-cyanopyridine 4-azabenzonitrile 4-cyanopyridine 4-pyridinenitrile 4-pyridyl cyanide isonicotinic acid nitrile isonicotinonitrile «gamma»-Cyanopyridine
<b>Inchi:</b>	InChI=1S/C6H4N2/c7-5-6-1-3-8-4-2-6/h1-4H
<b>InchiKey:</b>	GPHQHTOMRSGBNZ-UHFFFAOYSA-N
<b>Formula:</b>	C6H4N2
<b>SMILES:</b>	N#Cc1ccncc1
<b>Mol. weight [g/mol]:</b>	104.11
<b>CAS:</b>	100-48-1

## Physical Properties

Property code	Value	Unit	Source
affp	880.60	kJ/mol	NIST Webbook
basg	848.80	kJ/mol	NIST Webbook
chs	-3143.00 ± 0.50	kJ/mol	NIST Webbook
hf	283.50 ± 1.10	kJ/mol	NIST Webbook
hfs	210.30 ± 0.90	kJ/mol	NIST Webbook
hsub	73.20	kJ/mol	NIST Webbook
hsub	73.20 ± 0.60	kJ/mol	NIST Webbook
hsub	73.20 ± 0.60	kJ/mol	NIST Webbook
ie	10.70	eV	NIST Webbook
ie	10.40 ± 0.10	eV	NIST Webbook
ie	10.30	eV	NIST Webbook
log10ws	-1.59		Crippen Method
logp	0.953		Crippen Method
mcvol	83.000	ml/mol	McGowan Method
rinpol	955.00		NIST Webbook
rinpol	978.00		NIST Webbook
rinpol	955.00		NIST Webbook
ripol	1665.00		NIST Webbook
tb	352.00 ± 1.00	K	NIST Webbook
tf	352.40 ± 1.00	K	NIST Webbook

tf

353.55

K

Measurement and  
Correlation of Solubility of  
Two Isomers of  
Cyanopyridine in Eight  
Pure Solvents from 268.15  
K to 318.15 K

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C100481&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C100481&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Measurement and Correlation of Solubility of Two Isomers of Cyanopyridine in Eight Pure Solvents from 268.15 K to 318.15 K:</b>	<a href="https://www.doi.org/10.1021/acs.jced.7b00301">https://www.doi.org/10.1021/acs.jced.7b00301</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>chs:</b>	Standard solid enthalpy of combustion
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tf:</b>	Normal melting (fusion) point

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