

# p-menth-1-en-9-yl acetate

<b>Other names:</b>	p-mentha-1-en-9-yl acetate p-Ment-1-en-9-ol acetate p-Menth-1-en-9-ol, acetate
<b>Inchi:</b>	InChI=1S/C12H20O2/c1-9-4-6-12(7-5-9)10(2)8-14-11(3)13/h4,10,12H,5-8H2,1-3H3
<b>InchiKey:</b>	QUHIXSUMNSRNNP-UHFFFAOYSA-N
<b>Formula:</b>	C12H20O2
<b>SMILES:</b>	CC(=O)OCC(C)C1CC=C(C)CC1
<b>Mol. weight [g/mol]:</b>	196.29

## Physical Properties

Property code	Value	Unit	Source
gf	-141.42	kJ/mol	Joback Method
hf	-440.46	kJ/mol	Joback Method
hfus	18.77	kJ/mol	Joback Method
hvap	52.46	kJ/mol	Joback Method
log10ws	-2.97		Crippen Method
logp	2.932		Crippen Method
mcvol	172.220	ml/mol	McGowan Method
pc	2289.32	kPa	Joback Method
ripol	1410.00		NIST Webbook
ripol	1424.00		NIST Webbook
ripol	1410.00		NIST Webbook
ripol	1410.00		NIST Webbook
ripol	1791.00		NIST Webbook
ripol	1834.00		NIST Webbook
ripol	1828.00		NIST Webbook
ripol	1832.00		NIST Webbook
ripol	1824.00		NIST Webbook
ripol	1818.00		NIST Webbook
ripol	1834.00		NIST Webbook
ripol	1818.00		NIST Webbook
tb	573.50	K	Joback Method
tc	779.90	K	Joback Method
tf	302.82	K	Joback Method
vc	0.644	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	435.41	J/molxK	573.50	Joback Method
cpg	453.60	J/molxK	607.90	Joback Method
cpg	470.85	J/molxK	642.30	Joback Method
cpg	487.15	J/molxK	676.70	Joback Method
cpg	502.54	J/molxK	711.10	Joback Method
cpg	517.01	J/molxK	745.50	Joback Method
cpg	530.59	J/molxK	779.90	Joback Method
dvisc	0.0032213	Paxs	302.82	Joback Method
dvisc	0.0014754	Paxs	347.93	Joback Method
dvisc	0.0008084	Paxs	393.05	Joback Method
dvisc	0.0005014	Paxs	438.16	Joback Method
dvisc	0.0003399	Paxs	483.27	Joback Method
dvisc	0.0002463	Paxs	528.39	Joback Method
dvisc	0.0001877	Paxs	573.50	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R130142&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R130142&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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