

# Benzene, 2-chloro-1,3-dinitro-5-(trifluoromethyl)-

Other names:

Benzene, 2-chloro-5-(trifluoromethyl)-1,3-dinitro-  
Trifluoromethyl-3,5-dinitro-4-chlorobenzene  
3,5-Dinitro-4-chlorobenzotrifluoride  
4-Chloro-3,5-dinitrobenzotrifluoride  
Toluene, 4-chloro-«alpha», «alpha», «alpha»-trifluoro-3,5-dinitro-  
1-Chloro-2,6-dinitro-4-(trifluoromethyl)benzene  
2-Chloro-1,3-dinitro-5-(trifluoromethyl)benzene  
2,6-Dinitro-4-trifluoromethylchlorobenzene  
4-(Trifluoromethyl)-2,6-dinitrochlorobenzene  
4-Chloro-«alpha», «alpha», «alpha»-trifluoro-3,5-dinitrotoluene  
4-Chloro-3,5-dinitro-«alpha», «alpha», «alpha»-trifluorotoluene  
Toluene, 4-chloro-3,5-dinitro-«alpha», «alpha», «alpha»-trifluoro-  
Benzotrifluoride, 4-chloro-3,5-dinitro-  
3,5-Dinitro-4-chloro-«alpha», «alpha», «alpha»-trifluorotoluene  
Benzene, 1-chloro-2,6-dinitro-4-trifluoromethyl  
NSC 88274  
Chloralin

**Inchi:** InChI=1S/C7H2ClF3N2O4/c8-6-4(12(14)15)1-3(7(9,10)11)2-5(6)13(16)17/h1-2H  
**InchiKey:** HFHAVERN VFNSHL-UHFFFAOYSA-N  
**Formula:** C7H2ClF3N2O4  
**SMILES:** O=[N+]([O-])c1cc(C(F)(F)F)cc([N+](=O)[O-])c1Cl  
**Mol. weight [g/mol]:** 270.55  
**CAS:** 393-75-9

## Physical Properties

Property code	Value	Unit	Source
chs	-3087.00 ± 10.00	kJ/mol	NIST Webbook
gf	-430.84	kJ/mol	Joback Method
hf	-620.03	kJ/mol	Joback Method
hfs	-260.00 ± 0.88	kJ/mol	NIST Webbook
hfus	35.50	kJ/mol	Joback Method
hvap	69.26	kJ/mol	Joback Method
log10ws	-4.56		Crippen Method
logp	3.175		Crippen Method
mcvol	138.120	ml/mol	McGowan Method
pc	3329.68	kPa	Joback Method
tb	736.87	K	Joback Method

tc	991.26	K	Joback Method
tf	553.96	K	Joback Method
vc	0.576	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	342.87	J/mol×K	736.87	Joback Method
cpg	349.95	J/mol×K	779.27	Joback Method
cpg	356.27	J/mol×K	821.67	Joback Method
cpg	361.91	J/mol×K	864.07	Joback Method
cpg	366.94	J/mol×K	906.46	Joback Method
cpg	371.43	J/mol×K	948.86	Joback Method
cpg	375.45	J/mol×K	991.26	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C393759&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C393759&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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