

# Pregeijerene B

<b>Inchi:</b>	InChI=1S/C12H18/c1-11-7-4-3-5-8-12(2)10-6-9-11/h3-4,8-9H,5-7,10H2,1-2H3/b4-3-,11-9
<b>InchiKey:</b>	HJFJPKDXNZHKGG-DZTLHLKESA-N
<b>Formula:</b>	C12H18
<b>SMILES:</b>	CC1=CCCC(C)=CCC=CC1
<b>Mol. weight [g/mol]:</b>	162.27

## Physical Properties

Property code	Value	Unit	Source
gf	104.54	kJ/mol	Joback Method
hf	-90.59	kJ/mol	Joback Method
hfus	12.09	kJ/mol	Joback Method
hvap	45.93	kJ/mol	Joback Method
log10ws	-4.30		Crippen Method
logp	4.009		Crippen Method
mcvol	156.180	ml/mol	McGowan Method
pc	2619.09	kPa	Joback Method
rinpol	1274.00		NIST Webbook
rinpol	1274.00		NIST Webbook
rinpol	1274.00		NIST Webbook
ripol	1503.00		NIST Webbook
ripol	1503.00		NIST Webbook
ripol	1503.00		NIST Webbook
tb	522.70	K	Joback Method
tc	755.05	K	Joback Method
tf	249.86	K	Joback Method
vc	0.568	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	347.81	J/mol×K	522.70	Joback Method
cpg	368.58	J/mol×K	561.43	Joback Method
cpg	388.19	J/mol×K	600.15	Joback Method
cpg	406.64	J/mol×K	638.88	Joback Method

cpg	423.95	J/molxK	677.60	Joback Method
cpg	440.11	J/molxK	716.33	Joback Method
cpg	455.13	J/molxK	755.05	Joback Method
dvisc	0.0098765	Paxs	249.86	Joback Method
dvisc	0.0024293	Paxs	295.33	Joback Method
dvisc	0.0008688	Paxs	340.81	Joback Method
dvisc	0.0003958	Paxs	386.28	Joback Method
dvisc	0.0002128	Paxs	431.75	Joback Method
dvisc	0.0001288	Paxs	477.23	Joback Method
dvisc	0.0000850	Paxs	522.70	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R419628&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R419628&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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