

# Cyclopentene, 4,4-dimethyl-

<b>Other names:</b>	4,4-Dimethylcyclopentene
<b>Inchi:</b>	InChI=1S/C7H12/c1-7(2)5-3-4-6-7/h3-4H,5-6H2,1-2H3
<b>InchiKey:</b>	LKXAUCPURVBMRN-UHFFFAOYSA-N
<b>Formula:</b>	C7H12
<b>SMILES:</b>	CC1(C)CC=CC1
<b>Mol. weight [g/mol]:</b>	96.17
<b>CAS:</b>	19037-72-0

## Physical Properties

Property code	Value	Unit	Source
gf	69.08	kJ/mol	Joback Method
hf	-54.31	kJ/mol	Joback Method
hfus	2.75	kJ/mol	Joback Method
hvap	30.57	kJ/mol	Joback Method
log10ws	-2.26		Crippen Method
logp	2.363		Crippen Method
mcvol	94.330	ml/mol	McGowan Method
pc	3768.41	kPa	Joback Method
rinpol	696.00		NIST Webbook
tb	374.24	K	Joback Method
tc	579.63	K	Joback Method
tf	204.21	K	Joback Method
vc	0.352	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	162.74	J/molxK	374.24	Joback Method
cpg	177.82	J/molxK	408.47	Joback Method
cpg	191.74	J/molxK	442.70	Joback Method
cpg	204.60	J/molxK	476.93	Joback Method
cpg	216.48	J/molxK	511.16	Joback Method
cpg	227.50	J/molxK	545.39	Joback Method
cpg	237.75	J/molxK	579.63	Joback Method

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.29089e+01
Coeff. B	-2.66979e+03
Coeff. C	-5.08700e+01
Temperature range (K), min.	262.40
Temperature range (K), max.	402.28

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/files/research/kdb/mol/mol628.mol">https://www.thermo.com/files/research/kdb/mol/mol628.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C19037720&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C19037720&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpolar:</b>	Non-polar retention indices

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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