

# Isobutyl methyl phthalate

<b>Other names:</b>	1,2-Benzenedicarboxylic acid, isobutyl methyl ester
<b>Inchi:</b>	InChI=1S/C13H16O4/c1-9(2)8-17-13(15)11-7-5-4-6-10(11)12(14)16-3/h4-7,9H,8H2,1-3H
<b>InchiKey:</b>	HRJOMMSRROBCMQUHFFFAOYSA-N
<b>Formula:</b>	C13H16O4
<b>SMILES:</b>	<chem>COC(=O)c1ccccc1C(=O)OCC(C)C</chem>
<b>Mol. weight [g/mol]:</b>	236.26

## Physical Properties

Property code	Value	Unit	Source
gf	-308.92	kJ/mol	Joback Method
hf	-581.47	kJ/mol	Joback Method
hfus	25.13	kJ/mol	Joback Method
hvap	65.39	kJ/mol	Joback Method
log10ws	-2.97		Crippen Method
logp	2.286		Crippen Method
mcvol	185.150	ml/mol	McGowan Method
pc	2379.54	kPa	Joback Method
rinpol	1689.00		NIST Webbook
rinpol	1689.00		NIST Webbook
tb	680.64	K	Joback Method
tc	893.11	K	Joback Method
tf	404.53	K	Joback Method
vc	0.698	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	488.10	J/mol×K	680.64	Joback Method
cpg	502.29	J/mol×K	716.05	Joback Method
cpg	515.58	J/mol×K	751.46	Joback Method
cpg	527.98	J/mol×K	786.88	Joback Method
cpg	539.49	J/mol×K	822.29	Joback Method
cpg	550.12	J/mol×K	857.70	Joback Method
cpg	559.87	J/mol×K	893.11	Joback Method

dvisc	0.0012327	Paxs	404.53	Joback Method
dvisc	0.0006913	Paxs	450.55	Joback Method
dvisc	0.0004315	Paxs	496.57	Joback Method
dvisc	0.0002918	Paxs	542.59	Joback Method
dvisc	0.0002097	Paxs	588.60	Joback Method
dvisc	0.0001582	Paxs	634.62	Joback Method
dvisc	0.0001239	Paxs	680.64	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U373893&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U373893&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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