

# Cyclooctene, (Z)-

<b>Other names:</b>	cis-Cyclooctene (Z)-Cyclooctene c-Cyclooctene
<b>Inchi:</b>	InChI=1S/C8H14/c1-2-4-6-8-7-5-3-1/h1-2H,3-8H2/b2-1-
<b>InchiKey:</b>	URYYVOIYTNXXBN-UPHRSURJSA-N
<b>Formula:</b>	C8H14
<b>SMILES:</b>	C1=CCCCCCC1
<b>Mol. weight [g/mol]:</b>	110.20
<b>CAS:</b>	931-87-3

## Physical Properties

Property code	Value	Unit	Source
gf	54.40	kJ/mol	Joback Method
hf	-28.00 ± 1.90	kJ/mol	NIST Webbook
hf	-22.70	kJ/mol	NIST Webbook
hfus	4.26	kJ/mol	Joback Method
hvap	34.78	kJ/mol	Joback Method
ie	8.82	eV	NIST Webbook
log10ws	-2.92		Crippen Method
logp	2.897		Crippen Method
mcvol	108.420	ml/mol	McGowan Method
pc	3620.24	kPa	Joback Method
rinpol	895.00		NIST Webbook
rinpol	895.00		NIST Webbook
rinpol	883.00		NIST Webbook
rinpol	904.80		NIST Webbook
rinpol	913.00		NIST Webbook
tb	418.70	K	NIST Webbook
tb	418.00 ± 1.00	K	NIST Webbook
tc	635.63	K	Joback Method
tf	261.00 ± 3.00	K	NIST Webbook
tf	258.20 ± 3.00	K	NIST Webbook
vc	0.388	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	197.28	J/molxK	414.36	Joback Method
cpg	215.41	J/molxK	451.24	Joback Method
cpg	232.58	J/molxK	488.12	Joback Method
cpg	248.81	J/molxK	524.99	Joback Method
cpg	264.12	J/molxK	561.87	Joback Method
cpg	278.54	J/molxK	598.75	Joback Method
cpg	292.07	J/molxK	635.63	Joback Method
dvisc	0.0426132	Paxs	185.26	Joback Method
dvisc	0.0083526	Paxs	223.44	Joback Method
dvisc	0.0026344	Paxs	261.63	Joback Method
dvisc	0.0011148	Paxs	299.81	Joback Method
dvisc	0.0005729	Paxs	337.99	Joback Method
dvisc	0.0003370	Paxs	376.18	Joback Method
dvisc	0.0002186	Paxs	414.36	Joback Method
hvapt	41.57	kJ/mol	300.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C931873&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C931873&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature

<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinqol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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