

# 4,4'-Dimethoxydiphenylamine

**Other names:**

Bis(p-anisyl)amine  
Benzenamine, 4-methoxy-N-(4-methoxyphenyl)-  
4-Biphenylamine, 4,4'-dimethoxy-  
Bis(p-methoxyphenyl)amine  
Bis(4-methoxyphenyl)amine  
Di-p-anisylamine  
p,p'-Dimethoxydiphenylamine  
Di-p-methoxyphenylamine  
Termofleks A  
Di-(4-methoxy phenyl) amine  
4-Methoxy-N-(4-methoxyphenyl)benzenamine  
N-(p-Methoxyphenyl)-p-anisidine  
N,N-Bis(4-methoxyphenyl)amine  
N-(4-methoxyphenyl)-p-anisidine

**Inchi:**

InChI=1S/C14H15NO2/c1-16-13-7-3-11(4-8-13)15-12-5-9-14(17-2)10-6-12/h3-10,15H,1-

**InchiKey:**

VCOONNWIINSFBA-UHFFFAOYSA-N

**Formula:**

C14H15NO2

**SMILES:**

COc1ccc(Nc2ccc(OC)cc2)cc1

**Mol. weight [g/mol]:**

229.27

**CAS:**

101-70-2

## Physical Properties

Property code	Value	Unit	Source
gf	151.95	kJ/mol	Joback Method
hf	-93.14	kJ/mol	Joback Method
hfus	26.80	kJ/mol	Joback Method
hvap	63.89	kJ/mol	Joback Method
log10ws	-3.52		Crippen Method
logp	3.447		Crippen Method
mcvol	182.320	ml/mol	McGowan Method
pc	2627.15	kPa	Joback Method
tb	678.05	K	Joback Method
tc	910.42	K	Joback Method
tf	422.54	K	Joback Method
vc	0.674	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	474.82	J/molxK	678.05	Joback Method
cpg	490.45	J/molxK	716.78	Joback Method
cpg	504.97	J/molxK	755.51	Joback Method
cpg	518.40	J/molxK	794.23	Joback Method
cpg	530.77	J/molxK	832.96	Joback Method
cpg	542.09	J/molxK	871.69	Joback Method
cpg	552.38	J/molxK	910.42	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C101702&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C101702&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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