

# Benzenamine, 2,3,4,5,6-pentafluoro-

<b>Other names:</b>	Aniline, 2,3,4,5,6-pentafluoro- Pentafluoroaniline Pentafluorophenylamine 2,3,4,5,6-Pentafluoroaniline Aminopentafluorobenzene 1-Amino-2,3,4,5,6-pentafluorobenzene
<b>Inchi:</b>	InChI=1S/C6H2F5N/c7-1-2(8)4(10)6(12)5(11)3(1)9/h12H2
<b>InchiKey:</b>	NOXLGCOSAFGMDV-UHFFFAOYSA-N
<b>Formula:</b>	C6H2F5N
<b>SMILES:</b>	<chem>Nc1c(F)c(F)c(F)c(F)c1F</chem>
<b>Mol. weight [g/mol]:</b>	183.08
<b>CAS:</b>	771-60-8

## Physical Properties

Property code	Value	Unit	Source
gf	-843.70	kJ/mol	Joback Method
hf	-934.75	kJ/mol	Joback Method
hfus	23.99	kJ/mol	Joback Method
hvap	41.09	kJ/mol	Joback Method
ie	8.90	eV	NIST Webbook
ie	8.40 ± 0.02	eV	NIST Webbook
ie	8.60	eV	NIST Webbook
ie	8.90	eV	NIST Webbook
log10ws	-2.76		Crippen Method
logp	1.964		Crippen Method
mcvol	90.470	ml/mol	McGowan Method
pc	3435.91	kPa	Joback Method
ss	246.23	J/molxK	NIST Webbook
tb	426.70	K	NIST Webbook
tb	426.00	K	NIST Webbook
tc	638.12	K	Joback Method
tf	332.61	K	Joback Method
tt	306.75 ± 0.03	K	NIST Webbook
vc	0.383	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	224.64	J/mol×K	638.12	Joback Method
cpg	219.70	J/mol×K	607.96	Joback Method
cpg	191.58	J/mol×K	457.14	Joback Method
cpg	197.66	J/mol×K	487.30	Joback Method
cpg	203.50	J/mol×K	517.47	Joback Method
cpg	209.13	J/mol×K	547.63	Joback Method
cpg	214.52	J/mol×K	577.80	Joback Method
cps	230.79	J/mol×K	298.15	NIST Webbook
hfust	14.27	kJ/mol	306.80	NIST Webbook
hfust	3.94	kJ/mol	287.40	NIST Webbook
hfust	14.27	kJ/mol	306.80	NIST Webbook
sfust	46.51	J/mol×K	306.80	NIST Webbook
sfust	13.71	J/mol×K	287.40	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C771608&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C771608&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>ss:</b>	Solid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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