

# Butanal, 2-butenylhydrazone

<b>Other names:</b>	Butyraldehyde crotylhydrazone
<b>Inchi:</b>	InChI=1S/C8H16N2/c1-3-5-7-9-10-8-6-4-2/h3,5,8-9H,4,6-7H2,1-2H3/b5-3+,10-8?
<b>InchiKey:</b>	NJOLUQSSSLMAZEB-QXFJBYEUSA-N
<b>Formula:</b>	C8H16N2
<b>SMILES:</b>	CC=CCNN=CCCC
<b>Mol. weight [g/mol]:</b>	140.23
<b>CAS:</b>	36566-74-2

## Physical Properties

Property code	Value	Unit	Source
hf	44.46	kJ/mol	Joback Method
hvap	43.11	kJ/mol	Joback Method
ie	7.80 ± 0.10	eV	NIST Webbook
log10ws	-2.37		Crippen Method
logp	1.938		Crippen Method
mcvol	134.940	ml/mol	McGowan Method
pc	2436.25	kPa	Joback Method
tb	513.45	K	Joback Method
tc	708.74	K	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C36566742&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C36566742&amp;Units=SI</a>

## Legend

hf: Enthalpy of formation at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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