

# Heptane, 3-ethyl-

<b>Other names:</b>	3-Ethylheptane
<b>Inchi:</b>	InChI=1S/C9H20/c1-4-7-8-9(5-2)6-3/h9H,4-8H2,1-3H3
<b>InchiKey:</b>	PSVQKOKKLWHNRP-UHFFFAOYSA-N
<b>Formula:</b>	C9H20
<b>SMILES:</b>	CCCCC(CC)CC
<b>Mol. weight [g/mol]:</b>	128.26
<b>CAS:</b>	15869-80-4

## Physical Properties

Property code	Value	Unit	Source
af	0.4160		KDB
ap	346.150	K	KDB
gf	22.46	kJ/mol	Joback Method
hcg	6123.95	kJ/mol	KDB
hcn	5683.839	kJ/mol	KDB
hf	-234.37	kJ/mol	Joback Method
hfus	15.54	kJ/mol	Joback Method
hvap	44.50	kJ/mol	NIST Webbook
log10ws	-3.35		Crippen Method
logp	3.613		Crippen Method
mcvol	137.670	ml/mol	McGowan Method
pc	2330.00	kPa	KDB
rinpol	872.40		NIST Webbook
rinpol	871.10		NIST Webbook
rinpol	871.10		NIST Webbook
rinpol	867.40		NIST Webbook
rinpol	869.00		NIST Webbook
rinpol	869.00		NIST Webbook
rinpol	867.10		NIST Webbook
rinpol	864.00		NIST Webbook
rinpol	865.00		NIST Webbook
rinpol	865.00		NIST Webbook
rinpol	866.00		NIST Webbook
rinpol	867.00		NIST Webbook
rinpol	867.00		NIST Webbook
rinpol	868.00		NIST Webbook
rinpol	866.00		NIST Webbook

rinpol	867.00		NIST Webbook
rinpol	868.00		NIST Webbook
rinpol	868.00		NIST Webbook
rinpol	867.40		NIST Webbook
rinpol	870.30		NIST Webbook
rinpol	866.88		NIST Webbook
rinpol	869.57		NIST Webbook
rinpol	870.00		NIST Webbook
rinpol	866.60		NIST Webbook
rinpol	871.10		NIST Webbook
rinpol	870.50		NIST Webbook
rinpol	870.74		NIST Webbook
rinpol	871.27		NIST Webbook
rinpol	871.59		NIST Webbook
rinpol	870.44		NIST Webbook
rinpol	870.98		NIST Webbook
rinpol	871.37		NIST Webbook
rinpol	868.00		NIST Webbook
rinpol	870.00		NIST Webbook
rinpol	868.00		NIST Webbook
rinpol	871.00		NIST Webbook
rinpol	869.00		NIST Webbook
rinpol	871.00		NIST Webbook
rinpol	873.00		NIST Webbook
rinpol	867.00		NIST Webbook
rinpol	867.00		NIST Webbook
rinpol	867.00		NIST Webbook
rinpol	867.00		NIST Webbook
rinpol	867.00		NIST Webbook
rinpol	867.40		NIST Webbook
rinpol	865.00		NIST Webbook
rinpol	868.00		NIST Webbook
rinpol	870.00		NIST Webbook
rinpol	871.30		NIST Webbook
rinpol	870.00		NIST Webbook
tb	416.25 ± 0.40	K	NIST Webbook
tb	416.25 ± 0.20	K	NIST Webbook
tb	416.20	K	KDB
tb	416.75 ± 0.30	K	NIST Webbook
tc	587.50	K	KDB
tf	160.00	K	KDB
vc	0.533	m <sup>3</sup> /kmol	KDB
zc	0.2544750		KDB

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	339.92	J/molxK	543.79	Joback Method
cpg	327.48	J/molxK	516.01	Joback Method
cpg	314.53	J/molxK	488.22	Joback Method
cpg	301.08	J/molxK	460.44	Joback Method
cpg	287.11	J/molxK	432.66	Joback Method
cpg	272.60	J/molxK	404.88	Joback Method
cpg	351.89	J/molxK	571.57	Joback Method
dvisc	0.0108182	Paxs	176.19	Joback Method
dvisc	0.0002341	Paxs	404.88	Joback Method
dvisc	0.0003182	Paxs	366.76	Joback Method
dvisc	0.0004644	Paxs	328.65	Joback Method
dvisc	0.0007485	Paxs	290.53	Joback Method
dvisc	0.0013934	Paxs	252.42	Joback Method
dvisc	0.0032357	Paxs	214.31	Joback Method
hvapt	36.74	kJ/mol	416.20	KDB
rfi	1.40700		298.15	KDB

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.46094e+01
Coeff. B	-3.74920e+03
Coeff. C	-4.10930e+01
Temperature range (K), min.	302.88
Temperature range (K), max.	444.33

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	7.26259e+01
Coeff. B	-7.54929e+03
Coeff. C	-8.38301e+00

Coeff. D	4.01087e-06
Temperature range (K), min.	158.25
Temperature range (K), max.	590.00

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/files/research/kdb/mol/mol67.mol">https://www.thermo.com/files/research/kdb/mol/mol67.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C15869804&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C15869804&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=67">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=67</a>

## Legend

<b>af:</b>	Acentric Factor
<b>ap:</b>	Aniline Point
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hcg:</b>	Heat of Combustion, Gross form
<b>hcn:</b>	Heat of Combustion, Net Form
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rfi:</b>	Refractive Index
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume  
**zc:** Critical Compressibility

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