

Potassium hydroxide

Inchi: InChI=1S/K.H2O/h;1H2/q+1;/p-1
InchiKey: KWYUFKZDYNOTN-UHFFFAOYSA-M
Formula: H2K2O2
SMILES: [OH][K]
Mol. weight [g/mol]: 112.21
CAS: 12395-66-3

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.42925e+01
Coeff. B	-1.45074e+04
Coeff. C	-1.00390e+02
Temperature range (K), min.	679.00
Temperature range (K), max.	1600.00

Sources

Thermodynamic characteristics of reactions with KH(PO3H): Cheméo Relay (1.0, beta): <https://www.doi.org/10.1016/j.tca.2012.08.020>

Vapor-Liquid Equilibria of Ammonia + Water + Potassium Hydroxide and Ammonia + Water + Sodium Hydroxide Solutions for Calculating Phase Equilibria and Solubility of Hydroxide in NaOH/KOH Concentrated Electrolyte Solutions: Thermochemistry of Li, Na, K, Rb and Cs alkylated phenoxides: Density, viscosity, and N2O solubility of aqueous amino acid salt and amine liquid-liquid solutions of Aqueous Biphasic Systems Containing Selected Molar Volumes and Heat Capacities of Aqueous Solutions of Potassium Hydroxide and the System KOH-K2CrO4-H2O at 150 deg C in a High Alkali Concentrated Region: Capacities of the Ammonia + Water + NaOH and Ammonia + Water + KOH Solutions: <https://www.doi.org/10.1021/je049708+>
<https://www.doi.org/10.1016/j.fluid.2010.06.012>
<https://www.doi.org/10.1021/acs.jced.7b00690>
<https://www.doi.org/10.1016/j.tca.2005.02.027>
<https://www.doi.org/10.1016/j.jct.2011.09.012>
<https://www.doi.org/10.1021/je700315u>
<https://www.doi.org/10.1021/acs.jced.7b00192>
<https://www.doi.org/10.1021/je3004782>
<https://www.doi.org/10.1021/je050512z>
<http://webbook.nist.gov/cgi/cbook.cgi?ID=C12395663&Units=SI>

The Yaws Handbook of Vapor Pressure: Cheméo Relay (1.0): <https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

Vapor Pressure Determination of the KOH + K2CrO4 + H2O System: <https://www.doi.org/10.1021/je0501538>

Legend

pvap: Vapor pressure

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