

# Tribromochloromethane

<b>Inchi:</b>	InChI=1S/CBr3Cl/c2-1(3,4)5
<b>InchiKey:</b>	GKXZMEXQUWZGJK-UHFFFAOYSA-N
<b>Formula:</b>	CBr3Cl
<b>SMILES:</b>	ClC(Br)(Br)Br
<b>Mol. weight [g/mol]:</b>	287.18
<b>CAS:</b>	594-15-0

## Physical Properties

Property code	Value	Unit	Source
gf	-8.59	kJ/mol	Joback Method
hf	-9.47	kJ/mol	Joback Method
hfus	10.98	kJ/mol	Joback Method
hvap	40.21	kJ/mol	Joback Method
log10ws	-3.29		Crippen Method
logp	3.021		Crippen Method
mvol	89.690	ml/mol	McGowan Method
pc	7901.23	kPa	Joback Method
rinpol	955.00		NIST Webbook
tb	377.45 ± 0.60	K	NIST Webbook
tc	715.52	K	Joback Method
tf	312.77	K	Joback Method
vc	0.316	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	99.25	J/mol×K	454.96	Joback Method
cpg	101.63	J/mol×K	498.39	Joback Method
cpg	103.40	J/mol×K	541.81	Joback Method
cpg	104.65	J/mol×K	585.24	Joback Method
cpg	105.47	J/mol×K	628.67	Joback Method
cpg	105.95	J/mol×K	672.10	Joback Method
cpg	106.18	J/mol×K	715.52	Joback Method
dvisc	0.0029648	Paxs	312.77	Joback Method

dvisc	0.0020351	Paxs	336.47	Joback Method
dvisc	0.0014679	Paxs	360.17	Joback Method
dvisc	0.0011023	Paxs	383.87	Joback Method
dvisc	0.0008558	Paxs	407.56	Joback Method
dvisc	0.0006832	Paxs	431.26	Joback Method
dvisc	0.0005584	Paxs	454.96	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.61421e+01
Coeff. B	-4.34965e+03
Temperature range (K), min.	274.35
Temperature range (K), max.	401.64

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C594150&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C594150&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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