

# violanthrene-5,10-dione

<b>Other names:</b>	Dibenzanthrone (violanthrone)
<b>Inchi:</b>	InChI=1S/C34H16O2/c35-33-25-7-3-1-5-17(25)19-9-11-21-22-12-10-20-18-6-2-4-8-26(18)
<b>InchiKey:</b>	YKSGNOMLAIJTLT-UHFFFAOYSA-N
<b>Formula:</b>	C34H16O2
<b>SMILES:</b>	O=C1c2ccccc2-c2ccc3c4ccc5c6c(ccc(c7ccc1c2c73)c64)C(=O)c1ccccc1-5
<b>Mol. weight [g/mol]:</b>	456.49
<b>CAS:</b>	116-71-2

## Physical Properties

Property code	Value	Unit	Source
gf	846.94	kJ/mol	Joback Method
hf	515.61	kJ/mol	Joback Method
hfus	55.04	kJ/mol	Joback Method
hvap	118.24	kJ/mol	Joback Method
log10ws	-13.72		Crippen Method
logp	8.160		Crippen Method
mcvol	326.520	ml/mol	McGowan Method
pc	1687.95	kPa	Joback Method
tb	1311.78	K	Joback Method
tc	1614.52	K	Joback Method
tf	996.84	K	Joback Method
vc	1.296	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1365.28	J/molxK	1513.61	Joback Method
cpg	1428.68	J/molxK	1564.06	Joback Method
cpg	1171.86	J/molxK	1311.78	Joback Method
cpg	1212.38	J/molxK	1362.24	Joback Method
cpg	1257.73	J/molxK	1412.69	Joback Method
cpg	1308.49	J/molxK	1463.15	Joback Method
cpg	1499.29	J/molxK	1614.52	Joback Method
hsubt	208.80	kJ/mol	530.50	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C116712&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C116712&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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