

2,4,5-Trifluoro-3-methoxybenzoic acid, neopentyl ester

Inchi:	InChI=1S/C13H15F3O3/c1-13(2,3)6-19-12(17)7-5-8(14)10(16)11(18-4)9(7)15/h5H,6H2,1
InchiKey:	NPFGROSBKBLGQG-UHFFFAOYSA-N
Formula:	C13H15F3O3
SMILES:	COc1c(F)c(F)cc(C(=O)OCC(C)(C)C)c1F
Mol. weight [g/mol]:	276.25

Physical Properties

Property code	Value	Unit	Source
gf	-788.04	kJ/mol	Joback Method
hf	-1095.10	kJ/mol	Joback Method
hfus	27.71	kJ/mol	Joback Method
hvap	57.28	kJ/mol	Joback Method
log10ws	-4.26		Crippen Method
logp	3.315		Crippen Method
mcvol	188.890	ml/mol	McGowan Method
pc	1920.30	kPa	Joback Method
rinsol	1533.00		NIST Webbook
tb	636.73	K	Joback Method
tc	826.71	K	Joback Method
tf	411.35	K	Joback Method
vc	0.741	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	495.08	J/mol×K	636.73	Joback Method
cpg	508.45	J/mol×K	668.39	Joback Method
cpg	521.12	J/mol×K	700.06	Joback Method
cpg	533.11	J/mol×K	731.72	Joback Method
cpg	544.41	J/mol×K	763.39	Joback Method
cpg	555.03	J/mol×K	795.05	Joback Method
cpg	564.99	J/mol×K	826.71	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U357607&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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