

# Cyclopentane, chloro-

<b>Other names:</b>	CHLOROCYCLOPENTANE Cyclopentyl chloride
<b>Inchi:</b>	InChI=1S/C5H9Cl/c6-5-3-1-2-4-5/h5H,1-4H2
<b>InchiKey:</b>	NDTCXABJQNJPCF-UHFFFAOYSA-N
<b>Formula:</b>	C5H9Cl
<b>SMILES:</b>	C1CC1CC1
<b>Mol. weight [g/mol]:</b>	104.58
<b>CAS:</b>	930-28-9

## Physical Properties

Property code	Value	Unit	Source
gf	15.84	kJ/mol	Joback Method
hf	-101.79	kJ/mol	Joback Method
hfus	6.84	kJ/mol	Joback Method
hvap	38.79 ± 0.40	kJ/mol	NIST Webbook
hvap	38.80	kJ/mol	NIST Webbook
log10ws	-2.08		Crippen Method
logp	2.168		Crippen Method
mcpol	82.690	ml/mol	McGowan Method
pc	4135.61	kPa	Joback Method
rinpol	762.00		NIST Webbook
rinpol	804.00		NIST Webbook
rinpol	760.00		NIST Webbook
rinpol	760.00		NIST Webbook
rinpol	758.00		NIST Webbook
sl	238.04	J/mol×K	NIST Webbook
tb	387.20	K	NIST Webbook
tb	387.35 ± 0.20	K	NIST Webbook
tc	572.16	K	Joback Method
tf	178.85 ± 0.30	K	NIST Webbook
tt	180.00 ± 0.10	K	NIST Webbook
vc	0.305	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	140.78	J/molxK	400.78	Joback Method
cpg	183.28	J/molxK	537.88	Joback Method
cpg	152.33	J/molxK	435.06	Joback Method
cpg	192.44	J/molxK	572.16	Joback Method
cpg	163.25	J/molxK	469.33	Joback Method
cpg	173.56	J/molxK	503.61	Joback Method
cpg	128.57	J/molxK	366.51	Joback Method
cpl	152.40	J/molxK	298.15	NIST Webbook
dvisc	0.0004344	Paxs	336.58	Joback Method
dvisc	0.0005579	Paxs	306.65	Joback Method
dvisc	0.0011046	Paxs	246.79	Joback Method
dvisc	0.0017902	Paxs	216.86	Joback Method
dvisc	0.0033867	Paxs	186.93	Joback Method
dvisc	0.0007566	Paxs	276.72	Joback Method
dvisc	0.0003523	Paxs	366.51	Joback Method
hfust	7.63	kJ/mol	169.40	NIST Webbook
hfust	0.64	kJ/mol	180.00	NIST Webbook
hfust	0.64	kJ/mol	180.00	NIST Webbook
hvapt	37.40	kJ/mol	354.50	NIST Webbook
hvapt	38.79	kJ/mol	298.15	NIST Webbook
sfust	3.54	J/molxK	180.00	NIST Webbook
sfust	45.05	J/molxK	169.40	NIST Webbook
svapt	130.10	J/molxK	298.15	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.97435e+01
Coeff. B	-4.26454e+03
Coeff. C	-7.50310e+01
Temperature range (K), min.	294.22
Temperature range (K), max.	370.52

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	8.15692e+01
Coeff. B	-7.27943e+03
Coeff. C	-9.94014e+00
Coeff. D	7.40421e-06
Temperature range (K), min.	322.15
Temperature range (K), max.	386.15

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermopedia.com/doc/thermoprop/kdb/mol/mol1756.mol">https://www.thermopedia.com/doc/thermoprop/kdb/mol/mol1756.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C930289&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C930289&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.thermopedia.com/doc/thermoprop/kdb/hcprop/showprop.php?cmpid=1756">https://www.thermopedia.com/doc/thermoprop/kdb/hcprop/showprop.php?cmpid=1756</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemed.com/doc/models/crippen_log10ws">https://www.chemed.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices

<b>sfust:</b>	Entropy of fusion at a given temperature
<b>sl:</b>	Liquid phase molar entropy at standard conditions
<b>svapt:</b>	Entropy of vaporization at a given temperature
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/63-289-2/Cyclopentane-chloro.pdf>

Generated by Cheméo on 2024-04-20 05:49:23.602741413 +0000 UTC m=+15881412.523318739.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.