

# Benzoic acid, 2-(1-oxopropyl)-

<b>Other names:</b>	2-(1-oxopropyl)benzoic acid
<b>Inchi:</b>	InChI=1S/C10H10O3/c1-2-9(11)7-5-3-4-6-8(7)10(12)13/h3-6H,2H2,1H3,(H,12,13)
<b>InchiKey:</b>	JYBOOAWVGOYQFU-UHFFFAOYSA-N
<b>Formula:</b>	C10H10O3
<b>SMILES:</b>	CCC(=O)c1ccccc1C(=O)O
<b>Mol. weight [g/mol]:</b>	178.18
<b>CAS:</b>	2360-45-4

## Physical Properties

Property code	Value	Unit	Source
gf	-258.56	kJ/mol	Joback Method
hf	-402.06	kJ/mol	Joback Method
hfus	22.59	kJ/mol	Joback Method
hvap	70.96	kJ/mol	Joback Method
log10ws	-2.61		Crippen Method
logp	1.978		Crippen Method
mcvol	137.010	ml/mol	McGowan Method
pc	3699.95	kPa	Joback Method
tb	659.78	K	Joback Method
tc	867.95	K	Joback Method
tf	402.08	K	Joback Method
vc	0.518	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	338.64	J/molxK	659.78	Joback Method
cpg	381.81	J/molxK	833.25	Joback Method
cpg	374.37	J/molxK	798.56	Joback Method
cpg	366.36	J/molxK	763.86	Joback Method
cpg	357.75	J/molxK	729.17	Joback Method
cpg	348.52	J/molxK	694.47	Joback Method
cpg	388.70	J/molxK	867.95	Joback Method
dvisc	0.0000783	Paxs	659.78	Joback Method

dvisc	0.0001126	Paxs	616.83	Joback Method
dvisc	0.0001711	Paxs	573.88	Joback Method
dvisc	0.0002782	Paxs	530.93	Joback Method
dvisc	0.0004925	Paxs	487.98	Joback Method
dvisc	0.0009736	Paxs	445.03	Joback Method
dvisc	0.0022265	Paxs	402.08	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2360454&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2360454&amp;Units=SI</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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