

# Benzoic acid

## 2-ethoxymethyl-4,5-dimethoxy-tetrahydro-pyran-3-

Inchi:  
ester

InChI=1S/C17H24O6/c1-4-21-10-12-8-15(19-2)16(20-3)9-14(12)17(18)23-13-6-5-7-22-1

InchiKey:

FBWROJMKKNJULC-UHFFFAOYSA-N

Formula:

C17H24O6

SMILES:

CCOCc1cc(OC)c(OC)cc1C(=O)OC1CCCOC1

Mol. weight [g/mol]:

324.37

## Physical Properties

Property code	Value	Unit	Source
gf	-434.81	kJ/mol	Joback Method
hf	-911.23	kJ/mol	Joback Method
hfus	38.82	kJ/mol	Joback Method
hvap	79.02	kJ/mol	Joback Method
log10ws	-3.53		Crippen Method
logp	2.576		Crippen Method
mvol	246.690	ml/mol	McGowan Method
pc	1775.84	kPa	Joback Method
rinpol	2120.76		NIST Webbook
rinpol	2080.78		NIST Webbook
tb	820.03	K	Joback Method
tc	1037.10	K	Joback Method
tf	518.13	K	Joback Method
vc	0.911	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	768.63	J/molxK	820.03	Joback Method
cpg	785.18	J/molxK	856.21	Joback Method
cpg	800.27	J/molxK	892.39	Joback Method
cpg	813.89	J/molxK	928.57	Joback Method
cpg	826.00	J/molxK	964.75	Joback Method
cpg	836.59	J/molxK	1000.93	Joback Method
cpg	845.63	J/molxK	1037.10	Joback Method
dvisc	0.0003466	Paxs	518.13	Joback Method

dvisc	0.0002120	Paxs	568.45	Joback Method
dvisc	0.0001405	Paxs	618.76	Joback Method
dvisc	0.0000990	Paxs	669.08	Joback Method
dvisc	0.0000733	Paxs	719.40	Joback Method
dvisc	0.0000565	Paxs	769.71	Joback Method
dvisc	0.0000449	Paxs	820.03	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R273695&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R273695&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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