

# 3-Chlorodiphenylamine

<b>Other names:</b>	Benzenamine, 3-chloro-N-phenyl- Diphenylamine, 3-chloro-
<b>Inchi:</b>	InChI=1S/C12H10ClN/c13-10-5-4-8-12(9-10)14-11-6-2-1-3-7-11/h1-9,14H
<b>InchiKey:</b>	OHHIBZKYXJDQEU-UHFFFAOYSA-N
<b>Formula:</b>	C12H10ClN
<b>SMILES:</b>	Clc1cccc(Nc2ccccc2)c1
<b>Mol. weight [g/mol]:</b>	203.67
<b>CAS:</b>	101-17-7

## Physical Properties

Property code	Value	Unit	Source
gf	342.81	kJ/mol	Joback Method
hf	208.31	kJ/mol	Joback Method
hfus	23.82	kJ/mol	Joback Method
hvap	58.34	kJ/mol	Joback Method
log10ws	-3.96		Crippen Method
logp	4.084		Crippen Method
mvol	154.640	ml/mol	McGowan Method
pc	3276.53	kPa	Joback Method
rinpol	1789.60		NIST Webbook
rinpol	1789.60		NIST Webbook
tb	619.90	K	Joback Method
tc	872.29	K	Joback Method
tf	372.94	K	Joback Method
vc	0.576	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	351.71	J/mol×K	619.90	Joback Method
cpg	365.98	J/mol×K	661.96	Joback Method
cpg	379.03	J/mol×K	704.03	Joback Method
cpg	390.96	J/mol×K	746.09	Joback Method
cpg	401.84	J/mol×K	788.16	Joback Method

cpg	411.75	J/mol×K	830.22	Joback Method
cpg	420.77	J/mol×K	872.29	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	608.70	K	96.50	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C101177&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C101177&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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