

# Tridecane, 1-bromo-

<b>Other names:</b>	1-Bromotridecane Tridecyl bromide n-Tridecyl bromide n-Tridecyl-1-bromide
<b>Inchi:</b>	InChI=1S/C13H27Br/c1-2-3-4-5-6-7-8-9-10-11-12-13-14/h2-13H2,1H3
<b>InchiKey:</b>	BFDNZQUBFCYTIC-UHFFFAOYSA-N
<b>Formula:</b>	C13H27Br
<b>SMILES:</b>	CCCCCCCCCCCCBr
<b>Mol. weight [g/mol]:</b>	263.26
<b>CAS:</b>	765-09-3

## Physical Properties

Property code	Value	Unit	Source
gf	72.90	kJ/mol	Joback Method
hf	-285.32	kJ/mol	Joback Method
hfus	34.71	kJ/mol	Joback Method
hvap	50.97	kJ/mol	Joback Method
log10ws	-5.70		Crippen Method
logp	5.692		Crippen Method
mcvol	211.530	ml/mol	McGowan Method
pc	1740.46	kPa	Joback Method
rinpol	1652.00		NIST Webbook
rinpol	1658.00		NIST Webbook
rinpol	1651.00		NIST Webbook
rinpol	1652.00		NIST Webbook
rinpol	1651.00		NIST Webbook
rinpol	1658.00		NIST Webbook
ripol	1900.00		NIST Webbook
ripol	1905.00		NIST Webbook
tb	563.00	K	Joback Method
tc	736.25	K	Joback Method
tf	279.10 ± 0.30	K	NIST Webbook
vc	0.826	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	510.67	J/molxK	563.00	Joback Method
cpg	527.50	J/molxK	591.88	Joback Method
cpg	543.61	J/molxK	620.75	Joback Method
cpg	559.00	J/molxK	649.63	Joback Method
cpg	573.71	J/molxK	678.50	Joback Method
cpg	587.77	J/molxK	707.38	Joback Method
cpg	601.20	J/molxK	736.25	Joback Method
dvisc	0.0035921	Paxs	296.07	Joback Method
dvisc	0.0016097	Paxs	340.56	Joback Method
dvisc	0.0008683	Paxs	385.05	Joback Method
dvisc	0.0005323	Paxs	429.54	Joback Method
dvisc	0.0003577	Paxs	474.02	Joback Method
dvisc	0.0002573	Paxs	518.51	Joback Method
dvisc	0.0001950	Paxs	563.00	Joback Method
hvapt	64.60	kJ/mol	526.50	NIST Webbook

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	422.20	K	1.30	NIST Webbook
tbrp	421.00	K	1.30	NIST Webbook

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.59689e+01
Coeff. B	-5.23534e+03
Coeff. C	-9.82620e+01
Temperature range (K), min.	432.12
Temperature range (K), max.	589.50

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C765093&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C765093&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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