

# 2-Decene, (Z)-

<b>Other names:</b>	(Z)-2-Decene (Z)-C <sub>10</sub> H <sub>20</sub> cis-2-Decene
<b>Inchi:</b>	InChI=1S/C10H20/c1-3-5-7-9-10-8-6-4-2/h3,5H,4,6-10H2,1-2H3/b5-3-
<b>InchiKey:</b>	YKNMBTZOEVIJCM-HYXAFXHYSA-N
<b>Formula:</b>	C <sub>10</sub> H <sub>20</sub>
<b>SMILES:</b>	CC=CCCCCCCC
<b>Mol. weight [g/mol]:</b>	140.27
<b>CAS:</b>	20348-51-0

## Physical Properties

Property code	Value	Unit	Source
gf	113.54	kJ/mol	Joback Method
hf	-132.51	kJ/mol	Joback Method
hfus	21.86	kJ/mol	Joback Method
hvap	37.81	kJ/mol	Joback Method
ie	9.08 ± 0.01	eV	NIST Webbook
ie	8.90 ± 0.01	eV	NIST Webbook
log10ws	-3.86		Crippen Method
logp	3.923		Crippen Method
mcvol	147.460	ml/mol	McGowan Method
pc	2212.45	kPa	Joback Method
rinpol	1002.00		NIST Webbook
rinpol	1011.30		NIST Webbook
rinpol	1009.70		NIST Webbook
rinpol	1001.20		NIST Webbook
rinpol	1011.00		NIST Webbook
rinpol	1013.00		NIST Webbook
rinpol	1009.00		NIST Webbook
rinpol	1008.00		NIST Webbook
rinpol	1002.00		NIST Webbook
rinpol	1002.00		NIST Webbook
rinpol	1009.00		NIST Webbook
rinpol	1011.00		NIST Webbook
rinpol	1013.00		NIST Webbook
rinpol	1012.00		NIST Webbook
rinpol	1011.00		NIST Webbook

ripol	1010.00		NIST Webbook
ripol	1001.00		NIST Webbook
ripol	1011.30		NIST Webbook
ripol	1010.00		NIST Webbook
ripol	1002.00		NIST Webbook
ripol	1002.00		NIST Webbook
ripol	1012.00		NIST Webbook
ripol	1002.20		NIST Webbook
ripol	1081.00		NIST Webbook
ripol	1074.00		NIST Webbook
ripol	1078.00		NIST Webbook
ripol	1075.00		NIST Webbook
ripol	1076.00		NIST Webbook
ripol	1074.00		NIST Webbook
ripol	1080.70		NIST Webbook
ripol	1084.00		NIST Webbook
ripol	1073.70		NIST Webbook
ripol	1077.90		NIST Webbook
ripol	1080.70		NIST Webbook
ripol	1084.00		NIST Webbook
ripol	1073.70		NIST Webbook
ripol	1077.90		NIST Webbook
ripol	1072.00		NIST Webbook
ripol	1082.00		NIST Webbook
ripol	1084.00		NIST Webbook
ripol	1072.00		NIST Webbook
tb	432.36	K	Joback Method
tc	602.58	K	Joback Method
tf	197.38	K	Joback Method
vc	0.576	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	381.07	J/mol×K	602.58	Joback Method
cpg	300.30	J/mol×K	432.36	Joback Method
cpg	315.26	J/mol×K	460.73	Joback Method
cpg	329.59	J/mol×K	489.10	Joback Method
cpg	343.32	J/mol×K	517.47	Joback Method
cpg	356.46	J/mol×K	545.84	Joback Method
cpg	369.04	J/mol×K	574.21	Joback Method

dvisc	0.0001968	Paxs	432.36	Joback Method
dvisc	0.0057907	Paxs	197.38	Joback Method
dvisc	0.0020667	Paxs	236.54	Joback Method
dvisc	0.0009884	Paxs	275.71	Joback Method
dvisc	0.0005679	Paxs	314.87	Joback Method
dvisc	0.0003689	Paxs	354.03	Joback Method
dvisc	0.0002611	Paxs	393.20	Joback Method
hvapt	43.60	kJ/mol	424.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.74478e+01
Coeff. B	-4.76701e+03
Coeff. C	-6.84040e+01
Temperature range (K), min.	346.20
Temperature range (K), max.	461.20

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C20348510&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C20348510&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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