

# Carbonic acid, neopentyl 4-methoxyphenyl ester

Inchi:	InChI=1S/C13H18O4/c1-13(2,3)9-16-12(14)17-11-7-5-10(15-4)6-8-11/h5-8H,9H2,1-4H3
InchiKey:	SFIMQWQPJSPVFB-UHFFFAOYSA-N
Formula:	C13H18O4
SMILES:	COc1ccc(OC(=O)OCC(C)(C)C)cc1
Mol. weight [g/mol]:	238.28

## Physical Properties

Property code	Value	Unit	Source
gf	-279.72	kJ/mol	Joback Method
hf	-604.58	kJ/mol	Joback Method
hfus	20.83	kJ/mol	Joback Method
hvap	60.15	kJ/mol	Joback Method
log10ws	-3.40		Crippen Method
logp	3.257		Crippen Method
mvol	189.450	ml/mol	McGowan Method
pc	2227.09	kPa	Joback Method
rinpol	1695.00		NIST Webbook
rinpol	1695.00		NIST Webbook
tb	646.40	K	Joback Method
tc	857.34	K	Joback Method
tf	394.25	K	Joback Method
vc	0.705	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	498.95	J/molxK	646.40	Joback Method
cpg	514.56	J/molxK	681.56	Joback Method
cpg	529.23	J/molxK	716.71	Joback Method
cpg	542.97	J/molxK	751.87	Joback Method
cpg	555.78	J/molxK	787.03	Joback Method
cpg	567.69	J/molxK	822.19	Joback Method
cpg	578.71	J/molxK	857.34	Joback Method
dvisc	0.0009322	Paxs	394.25	Joback Method

dvisc	0.0005225	Paxs	436.27	Joback Method
dvisc	0.0003242	Paxs	478.30	Joback Method
dvisc	0.0002173	Paxs	520.33	Joback Method
dvisc	0.0001546	Paxs	562.35	Joback Method
dvisc	0.0001153	Paxs	604.38	Joback Method
dvisc	0.0000894	Paxs	646.40	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357862&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357862&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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