

D-Alanine, N-(4-ethylbenzoyl)-, dodecyl ester

Inchi:	InChI=1S/C24H39NO3/c1-4-6-7-8-9-10-11-12-13-14-19-28-24(27)20(3)25-23(26)22-17-1
InchiKey:	WDMIBDXWSLHAIR-UHFFFAOYSA-N
Formula:	C24H39NO3
SMILES:	CCCCCCCCCCCCOC(=O)C(C)NC(=O)c1ccc(CC)cc1
Mol. weight [g/mol]:	389.57

Physical Properties

Property code	Value	Unit	Source
gf	-21.91	kJ/mol	Joback Method
hf	-622.82	kJ/mol	Joback Method
hfus	57.53	kJ/mol	Joback Method
hvap	93.91	kJ/mol	Joback Method
log10ws	-7.45		Crippen Method
logp	5.831		Crippen Method
mvol	344.250	ml/mol	McGowan Method
pc	1056.88	kPa	Joback Method
rinpol	3029.00		NIST Webbook
tb	960.07	K	Joback Method
tc	1175.65	K	Joback Method
tf	558.93	K	Joback Method
vc	1.331	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1151.27	J/molxK	960.07	Joback Method
cpg	1168.12	J/molxK	996.00	Joback Method
cpg	1183.64	J/molxK	1031.93	Joback Method
cpg	1197.88	J/molxK	1067.86	Joback Method
cpg	1210.91	J/molxK	1103.79	Joback Method
cpg	1222.79	J/molxK	1139.72	Joback Method
cpg	1233.57	J/molxK	1175.65	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U354093&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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