

3-Methoxy-2,4,5-trifluorobenzoic acid, 4-biphenyl ester

Inchi:	InChI=1S/C20H13F3O3/c1-25-19-17(22)15(11-16(21)18(19)23)20(24)26-14-9-7-13(8-10)
InchiKey:	RWANLUGHJXIBLA-UHFFFAOYSA-N
Formula:	C20H13F3O3
SMILES:	COc1c(F)c(F)cc(C(=O)Oc2ccc(-c3ccccc3)cc2)c1F
Mol. weight [g/mol]:	358.31

Physical Properties

Property code	Value	Unit	Source
gf	-516.75	kJ/mol	Joback Method
hf	-769.24	kJ/mol	Joback Method
hfus	40.95	kJ/mol	Joback Method
hvap	79.37	kJ/mol	Joback Method
log10ws	-7.28		Crippen Method
logp	4.999		Crippen Method
mcvol	240.000	ml/mol	McGowan Method
pc	1848.33	kPa	Joback Method
rinpol	2654.00		NIST Webbook
tb	858.46	K	Joback Method
tc	1088.20	K	Joback Method
tf	553.18	K	Joback Method
vc	0.927	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	688.32	J/molxK	858.46	Joback Method
cpg	700.61	J/molxK	896.75	Joback Method
cpg	711.64	J/molxK	935.04	Joback Method
cpg	721.46	J/molxK	973.33	Joback Method
cpg	730.09	J/molxK	1011.62	Joback Method
cpg	737.56	J/molxK	1049.91	Joback Method
cpg	743.88	J/molxK	1088.20	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U360577&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvac:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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