

# 1-Pentene, 3,4-dimethyl-

<b>Other names:</b>	(CH <sub>3</sub> ) <sub>2</sub> CHCH(CH <sub>3</sub> )CH=CH <sub>2</sub> 3,4-Dimethyl-1-pentene
<b>Inchi:</b>	InChI=1S/C7H14/c1-5-7(4)6(2)3/h5-7H,1H2,2-4H3
<b>InchiKey:</b>	WFHXQNMTMDKVJG-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>7</sub> H <sub>14</sub>
<b>SMILES:</b>	C=CC(C)C(C)C
<b>Mol. weight [g/mol]:</b>	98.19
<b>CAS:</b>	7385-78-6

## Physical Properties

Property code	Value	Unit	Source
gf	91.02	kJ/mol	Joback Method
hf	-72.94	kJ/mol	Joback Method
hfus	5.56	kJ/mol	Joback Method
hvap	33.90	kJ/mol	NIST Webbook
log10ws	-2.12		Crippen Method
logp	2.465		Crippen Method
mcvol	105.190	ml/mol	McGowan Method
pc	2979.54	kPa	Joback Method
rinpol	636.80		NIST Webbook
rinpol	639.00		NIST Webbook
rinpol	641.30		NIST Webbook
rinpol	641.00		NIST Webbook
rinpol	642.00		NIST Webbook
rinpol	637.20		NIST Webbook
rinpol	647.00		NIST Webbook
rinpol	641.30		NIST Webbook
rinpol	634.60		NIST Webbook
rinpol	635.00		NIST Webbook
rinpol	644.20		NIST Webbook
rinpol	646.40		NIST Webbook
rinpol	637.20		NIST Webbook
rinpol	639.50		NIST Webbook
rinpol	642.00		NIST Webbook
rinpol	642.00		NIST Webbook
rinpol	638.00		NIST Webbook
rinpol	647.00		NIST Webbook

rinpol	637.30		NIST Webbook
rinpol	637.80		NIST Webbook
rinpol	637.00		NIST Webbook
rinpol	639.00		NIST Webbook
rinpol	638.00		NIST Webbook
rinpol	635.00		NIST Webbook
rinpol	638.00		NIST Webbook
rinpol	640.00		NIST Webbook
rinpol	642.00		NIST Webbook
rinpol	644.40		NIST Webbook
rinpol	636.40		NIST Webbook
rinpol	634.00		NIST Webbook
rinpol	638.00		NIST Webbook
rinpol	639.00		NIST Webbook
rinpol	634.00		NIST Webbook
rinpol	636.40		NIST Webbook
tb	353.95 ± 1.00	K	NIST Webbook
tb	363.05	K	NIST Webbook
tb	353.93 ± 0.30	K	NIST Webbook
tb	353.96 ± 0.20	K	NIST Webbook
tc	530.42	K	Joback Method
tf	136.89	K	Joback Method
vc	0.397	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	245.12	J/molxK	530.42	Joback Method
cpg	235.41	J/molxK	501.24	Joback Method
cpg	225.27	J/molxK	472.06	Joback Method
cpg	214.68	J/molxK	442.89	Joback Method
cpg	203.64	J/molxK	413.71	Joback Method
cpg	192.13	J/molxK	384.54	Joback Method
cpg	180.13	J/molxK	355.36	Joback Method
dvisc	0.0179569	Paxs	136.89	Joback Method
dvisc	0.0002201	Paxs	355.36	Joback Method
dvisc	0.0003016	Paxs	318.95	Joback Method
dvisc	0.0004481	Paxs	282.54	Joback Method
dvisc	0.0007487	Paxs	246.13	Joback Method
dvisc	0.0014946	Paxs	209.71	Joback Method
dvisc	0.0039899	Paxs	173.30	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.37935e+01
Coeff. B	-2.74141e+03
Coeff. C	-5.51640e+01
Temperature range (K), min.	258.14
Temperature range (K), max.	378.37

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=238">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=238</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7385786&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7385786&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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