

gallium

Inchi:	InChI=1S/Ga
InchiKey:	GYHNNYVSQQEPJS-UHFFFAOYSA-N
Formula:	Ga
SMILES:	[Ga]
Mol. weight [g/mol]:	69.72
CAS:	7440-55-3

Physical Properties

Property code	Value	Unit	Source
ea	0.43 ± 0.03	eV	NIST Webbook
ea	0.30 ± 0.15	eV	NIST Webbook
ea	0.40 ± 0.20	eV	NIST Webbook
ie	6.00	eV	NIST Webbook
ie	6.10 ± 0.40	eV	NIST Webbook
ie	6.10	eV	NIST Webbook
ie	6.00	eV	NIST Webbook
ie	6.00	eV	NIST Webbook
ie	6.00	eV	NIST Webbook
ie	6.00	eV	NIST Webbook
ie	6.00 ± 0.20	eV	NIST Webbook
ie	6.00 ± 0.30	eV	NIST Webbook
tf	302.92 ± 0.00	K	NIST Webbook
tf	303.40	K	Determination of thermophysical properties of cyclopentane hydrate using a stirred calorimetric cell
tt	302.92 ± 0.00	K	NIST Webbook
tt	302.92 ± 0.00	K	NIST Webbook

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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tcondl	29.30	W/m×K	321.30	A Novel Instrument for the Measurement of the Thermal Conductivity of Molten Metals. Part II: Measurements
tcondl	33.50	W/m×K	367.70	A Novel Instrument for the Measurement of the Thermal Conductivity of Molten Metals. Part II: Measurements
tcondl	37.00	W/m×K	416.30	A Novel Instrument for the Measurement of the Thermal Conductivity of Molten Metals. Part II: Measurements
tcondl	41.10	W/m×K	465.60	A Novel Instrument for the Measurement of the Thermal Conductivity of Molten Metals. Part II: Measurements
tcondl	45.00	W/m×K	520.40	A Novel Instrument for the Measurement of the Thermal Conductivity of Molten Metals. Part II: Measurements
tcondl	47.70	W/m×K	565.00	A Novel Instrument for the Measurement of the Thermal Conductivity of Molten Metals. Part II: Measurements
tcondl	50.90	W/m×K	614.30	A Novel Instrument for the Measurement of the Thermal Conductivity of Molten Metals. Part II: Measurements

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.69604e+01
Coeff. B	-3.09426e+04
Coeff. C	-1.37600e+01
Temperature range (K), min.	1310.15
Temperature range (K), max.	2518.15

Sources

Thermophysical Properties of the Liquid Ga-In-Sn Eutectic Alloy: Calorimetric study of enthalpies of mixing in liquid gallium germanium eutectic alloys

Experimental study on gallium activity in the liquid Ga-In-Tl alloys by EMF method with a zirconia solid electrolyte: NIST Webbook

Thermodynamic properties of Ga-Zn system. Experiment vs model: Thermodynamics of uranium in (Ga + Sn) eutectic alloy: Thermodynamic properties of liquid silver-gold-gallium alloys determined from Pb-Ag-Cu-Sn phase equilibria and liquid Ga-Zn alloys: Thermodynamic description of the Ga-Li-Zn system: Enthalpies of mixing of liquid Ag-Ga, Au-Ga and Ag-Au-Ga alloys: A Novel Instrument for the Measurement of the Thermal Conductivity of Molten Metals. Part II: Thermophysical properties of Ga-Zn: miscible alloys with Sn additions: Thermophysical properties of the liquid Ga-Sn-Zn eutectic alloy: Thermodynamic properties of liquid Ga-Li alloys: Experiment vs. modeling: Determination of thermophysical properties of cyclopentane hydrate vaporization in the Ga2O3-ZnO system by high temperature mass spectrometry: Partial and Integral Enthalpies of mixing of Ag-Ga-Sn liquid alloys: Mixing enthalpy of liquid Ga-Li-Zn alloys: Thermodynamic properties of ternary Me-Ga-In (Me = La, U) alloys in a fused Ga-In-Sn-Cu system: Vapor Pressure:

<https://www.doi.org/10.1021/je400882q>

<https://www.doi.org/10.1016/j.tca.2005.08.004>

<https://www.doi.org/10.1016/j.tca.2005.01.061>

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C7440553&Units=SI>

<https://www.doi.org/10.1016/j.jct.2016.07.029>

<https://www.doi.org/10.1016/j.jct.2015.09.023>

<https://www.doi.org/10.1016/j.tca.2018.02.012>

<https://www.doi.org/10.1016/j.fluid.2017.03.025>

<https://www.doi.org/10.1016/j.tca.2017.10.004>

<https://www.doi.org/10.1016/j.jct.2016.12.020>

<https://www.doi.org/10.1007/s10765-006-0057-y>

<https://www.doi.org/10.1016/j.fluid.2018.07.008>

<https://www.doi.org/10.1016/j.fluid.2018.03.001>

<https://www.doi.org/10.1016/j.jct.2016.02.015>

<https://www.doi.org/10.1016/j.jct.2018.05.023>

<https://www.doi.org/10.1016/j.jct.2017.07.009>

<https://www.doi.org/10.1016/j.tca.2011.04.032>

<https://www.doi.org/10.1016/j.tca.2015.12.011>

<https://www.doi.org/10.1016/j.jct.2018.10.014>

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

Legend

ea:	Electron affinity
ie:	Ionization energy
pvap:	Vapor pressure
tcondl:	Liquid thermal conductivity
tf:	Normal melting (fusion) point
tt:	Triple Point Temperature

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