

# Hexadecanoic acid, 1,2-ethanediyl ester

<b>Other names:</b>	Palmitic acid, ethylene ester Ethylene dipalmitate Ethylene glycol dihexadecanoate Ethylene glycol dipalmitate Ethylene palmitate 1,2-Di-O-Hexadecanoylethanol Ethanol, dihexadecanoic acid, ester 2-(Palmitoyloxy)ethyl palmitate ethane-1,2-diyl palmitate
<b>Inchi:</b>	InChI=1S/C34H66O4/c1-3-5-7-9-11-13-15-17-19-21-23-25-27-29-33(35)37-31-32-38-34(
<b>InchiKey:</b>	IKVCSHRLYCDSDU-UHFFFAOYSA-N
<b>Formula:</b>	C34H66O4
<b>SMILES:</b>	CCCCCCCCCCCCCCCC(=O)OCCOC(=O)CCCCCCCCCCCCCCCC
<b>Mol. weight [g/mol]:</b>	538.89
<b>CAS:</b>	624-03-3

## Physical Properties

Property code	Value	Unit	Source
gf	-232.44	kJ/mol	Joback Method
hf	-1234.69	kJ/mol	Joback Method
hfus	89.39	kJ/mol	Joback Method
hvap	109.59	kJ/mol	Joback Method
log10ws	-11.78		Crippen Method
logp	11.035		Crippen Method
mvol	504.800	ml/mol	McGowan Method
pc	518.88	kPa	Joback Method
rinpol	3656.00		NIST Webbook
tb	1129.90	K	Joback Method
tc	1472.40	K	Joback Method
tf	342.30 ± 0.50	K	NIST Webbook
vc	1.988	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1869.04	J/mol×K	1129.90	Joback Method
cpg	1897.14	J/mol×K	1186.98	Joback Method
cpg	1921.20	J/mol×K	1244.07	Joback Method
cpg	1941.52	J/mol×K	1301.15	Joback Method
cpg	1958.38	J/mol×K	1358.23	Joback Method
cpg	1972.09	J/mol×K	1415.31	Joback Method
cpg	1982.95	J/mol×K	1472.40	Joback Method
dvisc	0.0001351	Paxs	617.26	Joback Method
dvisc	0.0000585	Paxs	702.70	Joback Method
dvisc	0.0000304	Paxs	788.14	Joback Method
dvisc	0.0000179	Paxs	873.58	Joback Method
dvisc	0.0000116	Paxs	959.02	Joback Method
dvisc	0.0000081	Paxs	1044.46	Joback Method
dvisc	0.0000059	Paxs	1129.90	Joback Method

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C624033&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307i>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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