

# 2-(2-methylphenyl)-adamantane

<b>Inchi:</b>	InChI=1S/C17H22/c1-11-4-2-3-5-16(11)17-14-7-12-6-13(9-14)10-15(17)8-12/h2-5,12-15,
<b>InchiKey:</b>	VZOUAVQIZLTWJM-DSFJPSOSSA-N
<b>Formula:</b>	C17H22
<b>SMILES:</b>	Cc1ccccc1C1C2CC3CC(C2)CC1C3
<b>Mol. weight [g/mol]:</b>	226.36

## Physical Properties

Property code	Value	Unit	Source
gf	349.77	kJ/mol	Joback Method
hf	2.41	kJ/mol	Joback Method
hfus	27.89	kJ/mol	Joback Method
hvap	55.67	kJ/mol	Joback Method
log10ws	-4.78		Crippen Method
logp	4.535		Crippen Method
mcvol	194.050	ml/mol	McGowan Method
pc	2094.58	kPa	Joback Method
rinpol	1881.00		NIST Webbook
rinpol	1881.00		NIST Webbook
tb	635.17	K	Joback Method
tc	871.06	K	Joback Method
tf	362.11	K	Joback Method
vc	0.741	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	563.17	J/molxK	635.17	Joback Method
cpg	664.95	J/molxK	831.75	Joback Method
cpg	647.42	J/molxK	792.43	Joback Method
cpg	628.63	J/molxK	753.12	Joback Method
cpg	608.42	J/molxK	713.80	Joback Method
cpg	586.65	J/molxK	674.49	Joback Method
cpg	681.35	J/molxK	871.06	Joback Method
dvisc	0.0022041	Paxs	635.17	Joback Method

dvisc	0.0022213	Paxs	589.66	Joback Method
dvisc	0.0022415	Paxs	544.15	Joback Method
dvisc	0.0022656	Paxs	498.64	Joback Method
dvisc	0.0022949	Paxs	453.13	Joback Method
dvisc	0.0023312	Paxs	407.62	Joback Method
dvisc	0.0023775	Paxs	362.11	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R202178&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R202178&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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